

Tetrahedron

Tetrahedron 64 (2008) 4511-4574

www.elsevier.com/locate/tet

Tetrahedron report number 835

Advances in the Baylis—Hillman reaction-assisted synthesis of cyclic frameworks *

Vijay Singh, Sanjay Batra*

Medicinal and Process Chemistry Division, Central Drug Research Institute, PO Box 173, Lucknow 226001, India

Received 18 February 2008 Available online 29 February 2008

This article is dedicated to Professor D. Basavaiah for his seminal contributions toward development of this reaction

Keywords: Baylis-Hillman reaction; Cyclic compounds; Heterocycles; Nucleophilic substitution

Contents

1.	Introduction	4512
2.	Mechanism	4513
3.	Understanding opportunities for the generation of cyclic compounds based on the course of the reaction of the	
	Baylis-Hillman derivatives	4514
4.	Intramolecular Baylis—Hillman reaction	4515
5.	Intramolecular cyclization during the Baylis-Hillman reaction	4517
	5.1. Nitrogen-containing systems	
	5.2. Oxygen-containing systems	4519
	5.3. Sulfur-containing systems	
	5.4. Selenium-containing systems	4521
6.	Nucleophilic aromatic substitution reactions	452
7.	Friedel—Crafts reactions	4522

Abbreviations: Ac, acetyl; AIBN, 2,20-azobisisobutyronitrile; Aq, aqueous; Ar, aryl; 9BBN, 9-borabicyclo[3.3.1]nonane; BINAP, 2,20-bis(diphenylphosphanyl)-1,10-binaphthyl; BF3·Et₂O, boron trifluoride etherate; Bn, benzyl; Boc, tert-butoxycarbonyl; Bu, butyl; c, cyclo; CAN, ceric(IV) ammonium nitrate; cat., catalyst; Cp, cyclopentadienyl; DBU, 1,8-dazabicyclo[5,4.0]undec-7-ene; DCE, 1,2-dichloroethane; DDO, 2,3-dichloro-5,6-dicyanobenzoquinone; de, diastereomeric excess; DIBAL-H, diisobutylaluminum hydride; DIEA, diisopropylethylamine; DMAD, dimethyl acetylenedicarboxylate; DMAP, 4-dimethyl aminopyridine; DME, 1,2-dimethoxyethane; DMF, dimethylformamide; DMSO, dimethylsulfoxide; dppp, 1,3-bis(diphenylphosphino)propane; dr, diastereomeric ratio; DYKAT, dynamic asymmetric kinetic transformation; ee, enantiomeric excess; Et, ethyl; EtPh₂P, ethyldiphenylphosphine; EWG, electron-withdrawing group; Hex, hexyl; HFIP, hexafluoro-2-propanol; HFIPA, 1,1,1,3,3,3-hexafluoroisopropyl acrylate; HMDS, bis(trimethylsilyl)amide; HMP, hexamethyl phosphoramide; 3-HQD, 3-hydroxyquinuclidine; HMTA, hexamethylenetetramine; IBX, ο-iodoxybenzoic acid; β-ICD, β-isocupreidine; LDA, lithium diisopropylamide; m-CPBA, m-chloroperoxybenzoic acid; Me, methyl; Ms, mesyl; MOM, methoxymethyl; MS, molecular sieves; MVK, methyl vinyl ketone; MW, microwave; NBS, N-bromosuccinimide; NCS, N-chlorosuccinimide; NMO, N-methyl morpholine-N-oxide; Ns, p-nitrobenzenesulfonyl; Nu, Nucleophile; (o-Tol)₃P, tri-orthotolyl-phosphine; PCC, pyridinium chlorochromate; PEG, polyethylene glycol; Pent, pentyl; Ph, phenyl; PMB, p-methoxybenzoyl; PMDETA, N,N,N',N',N-pentamethyldiethylinetriamine; Mont., Montmorillonite; PPA, polyphosphoric acid; Pr, propyl; RCM, ring-closing metathesis; RDS, rate-determining step; rt, room temperature; SES, 2-trimethylsilylethylsulfonyl; t-Bu, tertiarybutyl; TBAF, tetrabutylammoniumfluoride; TBS or TBDMS, tert-butyldimethylsilyl; TBDPS, tert-butyldiphenylsilyl; TBHP, tetrabutylhydroperoxide; TFA, trifluoroacetic acid; TFAA, trifluoroacetic anhydride; TFE, trifluoroethanol; TfOH, triflic acid or trifluoromethanesulfonic acid; THBP, 2,4,5-trihydroxybutyrophenone; THF, tetrahydrofuran; TIPS, triisopropylsilyl; TMS, trimethylsilyl; TMSOTf, trimethylsilyl trifluoromethanesulfonate; Tol, toluene; TPAP, tetrapropylammonium perruthenate; Ts, tosyl or 4-toluenesulfonyl; TsOH, p-toluenesulfonic acid; TTMSS, tris(trimethylsilyl)silane; us, ultrasound.

[★] CDRI Communication No. 7357.

^{*} Corresponding author. Tel.: +91 522 2612411 18x4368; fax: +91 522 2623405. *E-mail address*: batra san@yahoo.co.uk (S. Batra).

8.	Cycloaddition reactions	4524
	8.1. 1,3-Dipolar cycloadditions	4524
	8.2. Diels-Alder reactions	4528
	8.3. (<i>n</i> + <i>n</i>) Annulations	4529
9.	Radical cyclizations	4531
10.	Heck coupling reactions	4534
11.	Ring-closing metathesis reactions	4536
12.	Epoxide formation	4540
13.	Aldol condensation	4540
14.	Claisen rearrangement	4541
15.	Lactonization reaction	4542
16.	Reductive cyclization	
	16.1. Reduction of aromatic nitro groups	
	16.2. Reduction of aliphatic nitro groups	
	16.2.1. Reduction of other functional groups	
17.	Hydrolysis of nitrile groups	
18.	Heterocyclization of substituted 3-aminopropanols and allyl amines	
19.	Ring transformations	4557
20.	Miscellaneous reactions	4558
21.	Natural products	4562
22.	Conclusions	4568
	Acknowledgements	
	References and notes	4570
	Biographical sketch	4574

1. Introduction

Acquiring the capability to access structurally complex and diverse molecules through simple starting substrates has been one of the underlying principles of chemical research. These diverse compounds are desired in order to serve mankind in a variety of ways. They might find use in pharmaceutical, agriculture, dyes, materials, electronics, and so forth. With the objective of generating an enormously complex skeletal diversity, chemists are always on the lookout for efficient complexity-generating reactions, also referred to as tandem reactions. These reactions may directly lead to a complex product from small and simple building blocks in a single operation or may lead to a product that is multifunctional and becomes a substrate for another complexity-generating reaction. Some of the examples of this class of reactions include the Ugi reaction, Passerini reaction, Diels-Alder reaction, ring-closing metathesis, and Baylis-Hillman reaction.

The Baylis—Hillman reaction, a carbon—carbon bond-forming reaction, which basically involves a reaction between an aldehyde (1) and an activated alkene (2) in the presence of a tertiary base, affords a highly functionalized product (3) (Fig. 1). As described by Basavaiah et al. in their recent review, it is a three-step reaction involving successive Michael,

Figure 1. General representation of the Baylis-Hillman reaction.

aldol, and elimination reactions in one pot.² The cheap and easy availability of starting materials, ease of performance (since it can be executed in water), atom economy, formation of chemospecific functional groups in the product, provision of an avenue for the introduction of asymmetry, and suitability for simulation on the solid phase as a prelude for combinatorial synthesis represent some of the reasons, which have led to an exponential increase in the synthetic utility of this reaction.

The reaction came into existence in 1968, when H. Morita disclosed that the reaction of an aldehyde with an activated alkene in the presence of tricvclohexylphosphine (PCv₃) affords a densely functionalized product (Fig. 2). Subsequently, he published a series of patents detailing the utility of his strategy.³ This reaction, however, earned its name from Baylis and Hillman, who reported that the reaction of aldehydes with activated alkenes including esters, amides, nitriles, and ketones in the presence of tertiary bicyclic amines furnished multifunctional products (Fig. 2). The synthetic appreciation of this reaction in organic chemistry was at a low ebb during the initial phase. When the utility of the Baylis-Hillman reaction for the synthesis of integerrinecic acid and mikanecic acid was, however, demonstrated by Drewes and Emslie⁵ in 1982 and Hoffmann and Rabe⁶ in 1983, respectively, various chemists became interested in it and started investigating its utility. This led to the publication of the first review in 1988 by Drewes and Roos. ⁷ Since then, investigations into different aspects of the Baylis-Hillman

Figure 2. Reactions embodied in patents by Morita and Baylis-Hillman.

reaction have gained pace and the second review on the subject, by Basavaiah et al., appeared in 1996, covering approximately 200 references.⁸ Simultaneously, Ciganek in 1997 authored a chapter on Organic Reactions, which discussed the Morita-Baylis—Hillman reaction in great detail. These articles fueled an explosive growth in the scope and applications of this reaction over the next 5 years, which was evident by the inclusion of almost 500 references in the review published by Basavaiah et al. in 2003. 10 In between these develoments, the progress of the asymmetric Baylis-Hillman reaction was separately reviewed by Langer¹¹ in 2000, while Kim et al. have assimilated the literature pertaining to the developments in the synthesis of cyclic compounds employing this reaction in 2002. 12 Since the publication of the review by Basavaiah et al. in 2003, more and more research groups have initiated work on different facets of this reaction. This has led to a search for increasing the scope of the substrates, for novel catalysts, for advancing an understanding of the mechanism and for the disclosure of a variety of synthetic applications of the derivatives generated from this reaction. Interestingly, a major portion of the synthetic applications relates to the synthesis of cyclic frameworks including the carbocyclic, heterocyclic, and benzannulated systems. Hence, we feel strongly that an assimilation of the literature since 2003 on this topic will be useful to organic chemists. It was a coincidince that, during the process of writing this report, three more reviews have appeared in the literature, which clearly reflects the relevance of this reaction at the present time. Besides the review on the Baylis-Hillman reaction by Basayaiah et al.. Shi et al.¹³ have reviewed the aza-Baylis-Hillman reaction, while Zhu et al. 14 have compiled the literature on the enantioselective Morita-Baylis-Hillman reaction and its aza counterpart. Because of the areas covered by these articles, we will be limiting this review to all references from 2003, which disclose the synthesis of cyclic systems employing the Baylis—Hillman adducts or their corresponding derivatives. References related to the chemistry originating from the Baylis—Hillman reaction of cyclic compounds (e.g., heterocyclic aldehydes and cyclic enones) have also been excluded. The literature covered in this article has been obtained either from a Sci-Finder search using the keyword 'Baylis—Hillman' or from other web resources. This review has been categorized on the basis of the reaction strategies involved in the process of cyclization. A section of miscellaneous reactions accommodates the strategies or isolated reports, which could not be classified under any of the mentioned sections. For the sake of greater attention, the applications relating to the generation of the key intermediates for the synthesis of natural products bearing a cyclic framework have been discussed at the end of the review.

2. Mechanism

The mechanism of the Baylis—Hillman reaction has already been discussed in detail by Basavaiah et al. 2,10 and Zhu et al. 14 The first acceptable mechanism for the Baylis—Hillman reaction was suggested by Hoffmann and Rabe 15 in 1983, although Morita has reported a mechanism in 1968 related to the phophine-mediated reactions. Later, based on pressure dependence, rate, and kinetic isotope effect data, Hill and Isaacs reported a mechanism, which was believed to proceed through a Michael-initiated addition—elimination sequence (Scheme 1). This was supported by Bode and Kaye through a rate law and was the most commonly accepted mechanism. Very recently, however, McQuade et al. have proposed a new mechanism involving a hemiacetal intermediate (I) (Scheme 2). This mechanism was based on the reaction rate data in aprotic solvents, where they determined that the Baylis—Hillman

Scheme 1.

RDS was second order in aldehyde and first order in DABCO and acrylate. Initially, they carried out studies with 4-nitrobenzaldehyde, but, later, they published their results with other aromatic aldehydes.¹⁹

More recently, Xu has carried out density function theory calculations for PMe₃-catalyzed Morita—Baylis—Hillman reactions with the objective of probing the mechanism. He concluded that the catalytic cycle of the Morita—Baylis—Hillman reaction involves four steps.²⁰ Initially, 1,4-addition of Me₃P to the activated alkene results in a zwitterionic intermediate followed by the zwitterionic phosphonium adduct attacking the ethanal. This is followed by an intramolecular hydrogen transfer and final elimination of the product from the intermediate (II) containing the product complexed to PMe₃, as shown in Scheme 3.

Scheme 3.

Roy and Sunoj reported the first DFT as well as ab initio investigations on the mechanism of the Morita—Baylis—Hillman reaction between (i) acrolein and formaldehyde catalyzed by trimethylamine (model system) and (ii) methyl vinyl ketone and benzaldehyde catalyzed by DABCO (real system).

Scheme 4.

They showed that the rate-limiting step involves an intramolecular proton transfer in the zwitterionic intermediate (III) afforded by the addition of enolate to electrophiles under polar aprotic conditions (Scheme 4). The activation barrier for the C–C bond formation was found to be lower than the proton-transfer step for the reaction of methyl vinyl ketone with benzaldehyde in the presence of DABCO.²¹

Subsequently, Robiette et al. carried out a detailed computational investigation for the mechanism of the Morita—Baylis—Hillman reaction. It was mentioned by the authors of this paper that the thorough calibration and mechanistic studies carried out by Roy and Sunoj partly mirror their work. They predicted, that in the absence of a protic solvent, deprotonation of the α -position is the rate-determining step and occurs through a cyclic transition state, with proton transfer to a hemiacetal alkoxide (IV) formed by the addition of a second equivalent of aldehyde to the intermediate alkoxide. On the contrary, in the presence of methanol they found a slightly lower energy pathway in which the alcohol serves as a carrier to transfer the proton from carbon to oxygen, as shown in Scheme 5. 22

3. Understanding opportunities for the generation of cyclic compounds based on the course of the reaction of the Baylis—Hillman derivatives

Any cyclization process requires the construction of a C-C or C-X bond, where X could be a heteroatom. The intramolecular Baylis—Hillman reaction offers the first opportunity for the synthesis of cyclic compounds. The Baylis—Hillman adducts incorporate three chemospecific groups, viz., a hydroxyl group, a double bond and an EWG. These groups could be appropriately tailored to generate an array of cyclic compounds directly from the Baylis—Hillman adducts. The Baylis—Hillman adduct could be readily transformed into the acetate (V) or allyl bromide (VI) via acetylation and bromination, respectively (Fig. 3). The acetate and allyl bromide

Figure 3. Structures of acetate (V) and allyl bromide (VI) derived from the Baylis—Hillman adduct.

Scheme 5.

undergo a variety of reactions, leading to products, which could be efficiently exploited for the generation of cyclic scaffolds. The pictorial depiction of different points of cyclization is delineated in Figure 4.

Since a large number of synthetic manipulations are initiated via the reactions of Baylis-Hillman adducts, and their acetyl or allyl bromide derivatives with nucleophiles, a short discussion on this subject is warranted. The nucleophilic attack on the Baylis-Hillman adducts mainly follows a Michael-type addition, in which the nucleophile adds to the carbon of the methylene group in the molecule, e.g., reaction of a primary amine yields the substituted 3-aminoalkanol VII (Fig. 5). In contrast, to this situation the nucleophilic substitution reaction of the Baylis-Hillman acetate depends upon the reaction conditions. If the nucleophile is added in an organic solvent in the presence of a base, an allylic substitution reaction is initiated, which proceeds sequentially via addition and elimination processes and is commonly referred as an S_N2'type reaction, e.g., reaction of a primary amine in an organic solvent furnishes the substituted allyl amine VIII (Fig. 6). When the same nucleophile is added to the Baylis-Hillman acetate in the presence of DABCO under aqueous conditions, however, an allylic substitution reaction results involving sequential steps such as addition of DABCO and elimination of acetic acid leading to an intermediate complex (IX). This

Figure 4. Different colors showing various opportunities for cyclization in the Baylis—Hillman adduct.

Figure 5. Nucleophilic addition in the Baylis—Hillman adduct, e.g., formation of substituted 3-aminopropanol VII.

Figure 6. $S_N 2'$ reaction of the nucleophile with the Baylis—Hillman acetate, e.g., formation of an allyl amine derivative **VIII**.

is followed by attack of the nucleophile on the double bond and elimination of the DABCO. This is generally referred as an S_N 2-type reaction and the course of the reaction is presented in Figure 7, e.g., S_N 2 reaction of a primary amine yielded the substituted allyl amine X.

4. Intramolecular Baylis-Hillman reaction

The intramolecular Baylis—Hillman reaction, which is an excellent strategy for the construction of a cyclic framework has gained significant progress since Murphy et al. reported it in 1997. Methot and Roush described the intramolecular vinylogous Morita—Baylis—Hillman reaction to access the central cyclopentane ring (4) of FR182877 (Scheme 6). Later, Koo et al. developed an efficient synthesis of diverse ω -formyl- α , β -unsaturated carbonyl compounds (5), which, under a Ph₃P-mediated intramolecular Baylis—Hillman reaction, yielded several biologically important polycyclic compounds (6), as shown in Scheme 7.

Scheme 6.

Krishna et al. reported the first diastereoselective intramolecular reaction of a chiral substrate in which both aldehyde and activated olefin co-exist as substituents to afford α -methylene- β -hydroxy- γ -butyrolactones (7) in good yield (Scheme 8). Further, they demonstrated the synthesis of a single isomer of tetrahydrofurano-2-pyrone (8), starting from 1,2-O-iso propylidine-3-O-acrylate- α -D-xylo-pentadialdo-1,4-furanose (Scheme 9). Subsequently, they attempted the synthesis of an α -methylene- β -hydroxylactam starting from a chiral acrylamide aldehyde, which, in turn, was derived from L-serine. They isolated 3-methylene-2-pyrrolone (9), however, instead of the expected product, as depicted in Scheme 10.

The intramolecular Morita—Baylis—Hillman reaction with a high level of enantioselectivity was accomplished under the influence of a co-catalyst system involving pipecolinic acid and *N*-methylimidazole by Miller et al.²⁷ They demonstrated this strategy for the synthesis of cycloalkenes (**10**). Replacement of pipecolinic acid with proline and its derivatives resulted in a low enantioselectivity (Scheme 11). Recently, Seidel and Gladysz reported that a chiral rhenium-containing phosphine (η^5 -C₅H₅)Re(NO)(Ph₃P)CH₂Ph₃P-catalyzed intramolecular Morita—Baylis—Hillman reaction in benzene or

Figure 7. S_N 2 reaction of the nucleophile with the Baylis-Hillman acetate in aqueous medium, e.g., formation of substituted allyl amine X.

Scheme 7.

Scheme 8.

Scheme 9.

Scheme 10.

Scheme 11.

Scheme 13.

as the electrophilic partner in an organocatalytic process was

12

chlorobenzene afforded carbocycles (11) in good yields, but with moderate enantioselectivity (Scheme 12).²⁸

Krische et al. demonstrated the feasibility of nucleophilic catalysis as a route to the enolate formation in Bu₃P/ (Ph₃P)₄Pd-promoted cross-coupling, resulting in cycloallylation of the mono-enone mono-allylic acetate to give the cyclic compound 12, as shown in Scheme 13. The transformation was accomplished by uniting the nucleophilic features of the Morita-Baylis-Hillman reaction with the electrophilic features of the Trost-Tsuji reaction via the above two-component catalyst systems.²⁹

The intramolecular variant of the Morita-Baylis-Hillman reaction encompassing leaving groups on an allylic framework described by Krafft and Haxell. They observed that the allylic chloride in the presence of Bu₃P provided a better yield of the cyclic enone (13) as compared to the allylic mesylate or allylic tosylate, as delineated in Scheme 14.³⁰ Subsequently, they extended their strategy to the intramolecular α-alkylation of enones using saturated alkyl halides under the influence of Me₃P for the synthesis of cyclic enones **14** (Scheme 15).³¹ Later, these workers, with the objective of studying the mechanism, also isolated for the first time a ketophosphonium salt (15), which is a Morita-Baylis-Hillman intermediate having trans geometry (Scheme 15).³² They suggested that electrostatic interaction was not the overriding electronic influence

Scheme 12. Scheme 14.

Scheme 15.

for defining the stereochemical outcome of the cyclization. They also demonstrated the success of their methodology in an aqueous medium in the presence of a phase-transfer catalyst.³³ The use of an epoxide as the electrophile in the Morita—Baylis—Hillman chemistry was demonstrated for the first time by these workers for the successful synthesis of various cyclic enones (**16**), as illustrated in Scheme 16.³⁴

Scheme 16.

Hong et al. for the first time reported an efficient prolinecatalyzed enantioselective intramolecular Baylis—Hillman reaction to obtain cyclohexenol derivatives (17). They observed that the addition of imidazole, as a co-catalyst, to the reaction mixture resulted in an inversion of selectivity of the product (18), as depicted in Scheme 17.³⁵

More recently, another intramolecular organometallic variation of the Morita—Baylis—Hillman reaction has been reported by Pigge et al.³⁶ It was demonstrated that *N*-benzyl acrylamide—ruthenium complexes **19** were transformed into the spirocyclic derivative (**20**) in the presence of Bu₃P and NaH (Scheme 18). The ruthenium-arene complex served as an electrophile, which was capable of trapping enolate generated in the reaction.

5. Intramolecular cyclization during the Baylis-Hillman reaction

Several nitrogen-, oxygen-, sulfur-, and selenium-containing heterocyclic systems have been afforded through intramolecular cyclizations, which take place concomitantly after the Baylis—Hillman reaction. Generally, such a property is shown by two structural classes of compounds. In one of these classes, there is a nitrogen atom present adjacent to the carbon bearing the formyl group, which participates in the intramolecular cyclization. In the second structural class, there is a nucleophilic moiety present on the carbon adjacent to the carbon

Scheme 18.

Scheme 19.

bearing the formyl unit. Once the Baylis—Hillman reaction has been accomplished, there is a tandem intramolecular cyclization via an attack of the nucleophile either on the double bond or on the electron-withdrawing group of the resulting Baylis—Hillman adducts, yielding a cyclic derivative.

5.1. Nitrogen-containing systems

Basavaiah and Rao reported the first example of electrophile induced Baylis—Hillman reaction wherein treatment of pyridine-2-carbaldehyde with alkyl vinyl ketones and cyclic enones under the influence of TMSOTf provided a one-pot synthesis of indolizines (21), as shown in Scheme 19.³⁷ Subsequently, they also demonstrated a facile synthesis of indolizine-fused chromones (22) from the reaction between the Baylis—Hillman adducts of pyridine-2-carbaldehyde and 1-benzopyran-4(4*H*)-ones, as presented in Scheme 20.³⁸

Almeida et al. reported the Baylis—Hillman reaction of chiral α -amino aldehydes under sonication which avoids racemization and described the synthesis of bicyclic lactam (23) with an indolizidinic skeleton from the Baylis—Hillman adduct of *N*-Boc-pipecolaldehyde (Scheme 21).³⁹ Very recently, in an alternate strategy for the generation of *N*-bridgehead heterocyclic systems, Clive et al. reported the synthesis of bicyclic amines with nitrogen at a ring-fusion position (24) via a sequential Baylis—Hillman reaction between *N*-protected β -amino aldehydes and acrylates followed by O-acetylation and N-deprotection, as shown in Scheme 22.⁴⁰ They extended this strategy for the synthesis of (—)- δ -coniceine.

Virieux et al. have described the synthesis of indolizine (25) using the Baylis—Hillman reaction of pyrrolecarbaldehyde with an electron-deficient allene in the presence of a catalytic amount of Bu_3P (Scheme 23).⁴¹

Lee et al. demonstrated that the Baylis—Hillman reaction of 2-cyano benzaldehyde with activated alkenes in the presence of DABCO led to the formation of 2-(3-oxo-2,3-dihydro-1*H*-

Scheme 20.

Scheme 21.

Scheme 22.

Scheme 23.

isoindol-1-yl)-acrylates (26). When acrylonitrile and phenyl vinyl sulfone were employed as the alkenes, however, the rearranged products (27) were obtained, as illustrated in Scheme 24. Later Lee's group successfully accomplished the synthesis of several 1,2-dihydroquinoline derivatives (28) and 3-acetoxymethyl-quinolines (29) in moderate-to-good yields by the reaction between the acetyl derivatives of Baylis—Hillman

adducts of 2-azidobenzaldehydes with triethyl phosphite (Scheme 25).⁴³ Dong et al. have described the synthesis of substituted indole *N*-oxides (**30**) via a TiCl₄-mediated Baylis—Hillman reaction of α -oxo cyclic ketene-*S*,*S*-acetal (**31**) with 2-nitrobenzaldehydes (Scheme 26).⁴⁴

Scheme 26.

Scheme 24.

Scheme 25.

5.2. Oxygen-containing systems

Kim et al. demonstrated that the DABCO-catalyzed Baylis—Hillman reaction of 2-carboxybenzaldehyde with activated alkenes gave the enol lactones (32) in moderate yields via a sequential Baylis—Hillman reaction, and lactonization followed by a 1,3-hydrogen shift (Scheme 27).⁴⁵

Kaye et al. successfully developed a route to access 3-(chloromethyl) coumarin (33) via the direct cyclization of unprotected Baylis—Hillman adducts. ⁴⁶ Subsequently, Kaye and Musa demonstrated a general chemoselective approach to 3-substituted coumarins (34) from the Baylis—Hillman adducts of *O*-benzyl-protected salicylaldehydes via Pd—C-mediated hydrogenation, as delineated in Scheme 28. ⁴⁷

Kim et al. described the synthesis of chromenones (35) from the Baylis—Hillman reaction between salicylaldehydes and cycloalkenones under the influence of DMAP. They reported that the use of DABCO gave the products in low yields (Scheme 29). Subsequently, Lesch and Brase successfully achieved the synthesis of tetrahydroxanthenones (36) by carrying out the reaction in the presence of DABCO under sonication (Scheme 30). Later, they transformed these tetrahydroxanthenones into highly functionalized tetrahydroxanthenols (37, 38, and 39) (Scheme 31). In subsequent studies, however, in which they proposed the mechanistic details, they found their reaction to be an oxa-Michael—aldol condensation, rather than a Baylis—Hillman reaction.

Shi and Shi have also described a successful synthesis of tetrahydroxanthenes (40) from the Baylis—Hillman reaction of salicyl *N*-tosylimines with 2-cylohexenone in the presence

Scheme 30.

of PPhMe₂ (Scheme 32).⁵² They reported a similar reaction with cyclopentenone to afford the cyclized product (**41**) and this was dependent upon the nature of the salicyl *N*-tosylimine.

Shi et al. reported an efficient approach to highly functionalized chromenes (**42**) via a DABCO-catalyzed reaction of salicyl *N*-tosylimines with ethyl 2-butynoate and penta-3,4-diene-2-one in dichloromethane.⁵³ It was observed that the phosphine catalysts were not suitable for this particular reaction as they resulted in [3+2] cycloadducts only. These workers proposed that the formation of chromenes under an amine-catalyzed pathway proceeds via the formation of an unstable zwitterionic intermediate followed by a Michael/Mannich reaction. Based on this rationale they extended their study by reacting ethyl 2-butynoate and salicyl *N*-tosylimine. Unfortunately, the resulting chromene (**43**) was obtained in low yield, possibly due to the steric hindrance in ethyl 2-butynoate toward the nucleophilic attack (Scheme 33).

Shi et al. later described that, compared to ethyl 2-buty-noate, the reaction between diethyl acetylenedicarboxylate and salicyl *N*-tosylimine in the presence of DABCO or PPhMe₂ in DMSO proceeded efficiently to afford the corresponding chromenes (**44**) in excellent yields, as shown in Scheme 34.⁵⁴ Subsequently, Shi and Shi prepared substituted

Scheme 27.

$$R_{1} = R_{1} = R_{1} = R_{1} = R_{1} = R_{2} = R_{1} = R_{2} = R_{2$$

Scheme 28.

Scheme 29.

Scheme 31.

Scheme 32.

NHTs
$$CO_2Et$$
 CO_2Et CO_2E

Scheme 33.

Scheme 34.

chromenes (**45**) by the reaction of but-3-yn-2-one and methyl propiolate with salicyl *N*-tosylimines (Scheme 35). On the basis of their proton NMR observations, the reaction mechanism for this transformation was also delineated. Very recently, Qi and Shi reported the DABCO-catalyzed aza-Baylis—Hillman reaction of salicyl *N*-tosylimines with methyl vinyl ketone,

Scheme 35.

ethyl vinyl ketone or phenyl vinyl ketone.⁵⁶ They found that performing the reaction in toluene afforded chromans (**46**) as the major products, whereas a similar reaction in THF led to the formation of the aza-Baylis—Hillman adducts (**47**) as the major products (Scheme 36).

Recently, an interesting reaction describing the formation of 1,3-dioxolanes (**48** and **49**) during the DBU-catalyzed Baylis—Hillman coupling of a sesquiterpene lactone, parthenin, with aromatic and aliphatic aldehydes has been reported by Taneja et al. (Scheme 37).⁵⁷ They proposed that the presence of a tertiary homoallylic hydroxyl group at the C-1 position may be responsible for the 1,3-dioxolane formation.

5.3. Sulfur-containing systems

Inspired by the work of Brase et al. on the preparation of tetrahydroxanthenones, ^{47–49} Cardova et al. transformed 2-mercaptobenzaldehyde in the presence of (*S*)-prolinol, an asymmetric organocatalyst, into tetrahydrothioxanthanes (**50**). ⁵⁸ The Baylis—Hillman-type product was, in fact, obtained via a domino thia-Michael/aldol reaction (Scheme 38).

TsHN O TsHN O DABCO
$$R_4$$
 R_4 R_5 R_4 R_5 R_6 R_7 R_8 R_8 R_8 R_8 R_9 $R_$

Scheme 37.

Scheme 38.

Scheme 39.

Subsequently, these also generated thiobenzopyrans (**51**) in the presence of a chiral diarylprolinol, as delineated in Scheme 39.⁵⁹ Wang et al. independently studied other organocatalysts such as diarylprolinol silyl ethers and cinchona alkaloid-derived thioureas for the reaction of 2-mercaptoaldehydes, resulting in similar products **52** and **53** with good-to-excellent yields and enantioselectivities, as shown in Schemes 40 and 41, respectively.⁶⁰

Scheme 42.

5.4. Selenium-containing systems

Kataoka et al. have described the Baylis—Hillman reaction of 1-[2-(methylchalcogeno)phenyl]propenones with aldehydes, and found that seleno-substituted chalcogen afforded an unusual product, selenochromanone (**54**), albeit in low yield, along with the usual Baylis—Hillman product (Scheme 42).⁶¹

6. Nucleophilic aromatic substitution reactions

The nucleophilic aromatic substitution reaction was first introduced by Kim et al. for the synthesis of quinolines. Later, this group extended their strategy to the synthesis of *N*-substituted 1,4-dihydroquinolines and 2-substituted naphthalenes. They observed that the electron-withdrawing

$$\begin{array}{c} O \\ R \\ \hline \\ \hline \\ R \\ \hline \\ R \\ \hline \\ SH \\ R' \\ \\ \hline \\ R' \\ \\ \hline \\ CHO \\ \hline \\ PhMe, rt, MS 4 \\ A \\ \hline \\ 72-97\%, 85-95\% \\ ee \\ \hline \\ R' \\ \hline \\ \\ SR' \\ \hline \\ R' \\ \hline \\ CHO \\ Ar \\ H \\ Cat. \\ OSiMe_3 \\ R' \\ \hline \\ R' \\ \hline \\ CHO \\ Ar \\ H \\ Cat. \\ OSiMe_3 \\ R' \\ \hline \\ R' \\ \hline \\ R' \\ \hline \\ R' \\ \hline \\ CHO \\ Ar \\ H \\ Cat. \\ OSiMe_3 \\ R' \\ \hline \\ R' \\ \hline \\ R' \\ \hline \\ CHO \\ Ar \\ H \\ Cat. \\ OSiMe_3 \\ R' \\ \hline \\ R' \\ \hline \\ CHO \\ Cat. \\ OSiMe_3 \\ R' \\ \hline \\ R' \\ \hline \\ Cat. \\ OSiMe_3 \\ R' \\ \hline \\ R' \\ \hline \\ Cat. \\ OSiMe_3 \\ R' \\ \hline \\ Cho \\ Cat. \\ OSiMe_3 \\ R' \\ \hline \\ Cho \\ Cat. \\ OSiMe_3 \\ R' \\ \hline \\ Cho \\ Ch$$

Scheme 40.

 $R' = Ph, \ 2 - Cl - C_6H_4, \ 3 - Cl - C_6H_4, \ 4 - Cl - C_6H_4, \ 4 - MeO - C_6H_4, \ 3 - Me - C_6H_4, \ 1 - naphthyl; \ Ar = 3,6 - (CF_3)_2C_6H_3$

conjugated ester moiety is crucial to render the $S_N\!Ar$ reaction more facile.

Hong and Lee reported the synthesis of coumarins (55) via intramolecular aromatic substitution followed by rearrangement of 3-(2-chloro-phenyl)-2-nitromethyl-acrylic acid methyl esters afforded from the reaction between the Baylis-Hillman acetates and NaNO₂ (Scheme 43).⁶⁴ Subsequently Horn and Perez disclosed the formation of 3-nitroindole derivatives (56) from the reaction between the Baylis-Hillman acetates and KNO₂.65 It was presumed that the 2-nitromethyl-3-(2nitro-phenyl)-acrylic acid methyl esters, initially generated, cyclized via nucleophilic aromatic substitution to yield the indole derivatives, as shown in Scheme 43. Recently, Hong and Lee have demonstrated that the treatment of methyl propiolate with substituted 2-fluorobenzaldehydes in the presence of ZrCl₄/Bu₄NI offers the β-iodo-α-(hydroxyalkyl)acrylates, which, upon reaction with primary amines, lead to quinolones (57), according to the reaction sequence shown in Scheme 44.66

The acetyl derivatives of Baylis—Hillman adducts, derived from 2-chloronicotinal dehydes, were successfully utilized for the development of a convenient protocol for the general synthesis of 8-methyl- and 8-cyano-quinolines (**58** and **59**) by Rao et al. ⁶⁷ The Baylis—Hillman acetates by undergoing an S_N2' reaction with nitroethane or ethyl cyanoacetate via a successful S_N2 — S_NAr elimination led to these quinolines (**58** and **59**), which exhibited substantial antibacterial and antifungal activities (Scheme 45).

7. Friedel-Crafts reactions

The use of the Friedel-Crafts reaction in the Baylis-Hillman chemistry was initially introduced by Basavaiah during a stereoselective synthesis of trisubstituted alkenes⁶⁸ and was later described by several research groups. 69 Basavaiah et al. described the synthesis of 2-benzazepines (60) via C-N bond formation between the Baylis-Hillman adducts and nitriles in the presence of methylsulfonic acid. 70 Subsequently, they also reported the synthesis of 2-benzoxepines (61) via C-O bond formation between the Baylis-Hillman adducts and HCHO in the presence of sulfuric acid (Scheme 46).⁷¹ Recently, Das et al. demonstrated the synthesis of similar 2-benzoxepines using the same substrates in the presence of silica-supported hypochlorite.⁷² Interestingly, Kim et al. developed the synthesis of 7H-benzocycloheptene derivatives (62), which are carbon surrogates of 2-benzazepines.⁷³ Their preparation involved an intramolecular Friedel-Crafts alkenylation reaction with an electron-deficient arene unit, as shown in Scheme 47.

Kim et al. have utilized the Baylis—Hillman acetates for the synthesis of indeno[1,2-b]quinolin-10-ones (**63** and **64**). The S_N2' reaction of anilines with Baylis—Hillman acetates afforded the allyl amines, which, upon treatment with PPA, underwent a double Friedel—Crafts reaction to yield the tetrahydro-indeno[1,2-b]quinolones (**63**) as the major products along with minor yields of the 7H-indeno[2,1-c]quinolines (**64**) (Scheme 48). Later, the same workers showed that

Scheme 43.

Scheme 44.

RO₂C
$$R_2$$
 EtNO₂, K₂CO₃ R_2 R_2 R_3 R_4 R_5 R_5 R_6 R_6 R_6 R_7 R_8 R_8 R_8 R_8 R_8 R_9 $R_$

Scheme 45.

R₂ CO₂R₁ HCHO, conc. H₂SO₄
$$R_2$$
 CH₂Cl₂, rt, 1 h R_2 HCHO, conc. H₂SO₄ R_3 R₄ R₁ = Me, Et R_2 H, Me; R₃ = H, Me, Et, i-Pr; R₄ = H R_3 R = Me, Et; R₂ = R₃ = R₄ = OMe, OPr **60**

Scheme 46.

$$\begin{array}{c} R_{2} & \text{OH} \\ \text{CO}_{2}\text{Me} & \begin{array}{c} \text{PhCCMgBr} \\ \text{CuI, THF, rt, 8 h} \\ \text{46-75\%} & R_{1} \end{array} \\ \begin{array}{c} R_{2} \\ \text{O °C to rt} \end{array} \\ \begin{array}{c} R_{2} \\ \text{O °C to rt} \end{array}$$

Scheme 47.

OAC
$$CO_2Me$$
 H_2N R_2 THF , reflux, 14-18 h R_1 R_2 R_2 R_3 R_4 R_5 R_6 R_7 R_8 R_8 R_9 R_9

Scheme 48.

OAc CO₂Me CO₂Me
$$\frac{1. \text{ phthalimide}}{2. \text{ NaBH}_4}$$
 R $\frac{1. \text{ phthalimide}}{1. \text{ phthalimide}}$ R $\frac{1.$

Scheme 49.

phthalimide-substituted Baylis—Hillman derivatives (**65**), on treatment with methylsulfonic acid in dichloroethane, afforded the benzo[3,4]azepino[2,1-*a*]isoindoles (**66**) in excellent yields (Scheme 49).⁷⁵ Shanmugam et al. developed an efficient and eco-friendly protocol for the synthesis of indene derivatives (**67**) from the Baylis—Hillman adducts via a Mont. K-10-mediated intramolecular Friedel—Crafts reaction (Scheme 50).⁷⁶

Lee et al. described a simple protocol for the synthesis of 1-hydroxynaphthalenes (68) from the allyl cyanides, which, in turn, were obtained from the reaction between the Baylis—Hillman acetates and KCN in the presence of DMSO (Scheme

OH
$$CO_2$$
Me $\frac{2\text{-propynyl alcohol},}{\text{Mont. K-10, 80 °C, 12 h}}$ R_1O R_2O $R_2 = Me$, Et, -CH₂- R_2O $R_3 = R_2 = Me$, Et, -CH₂- R_2O $R_3 = R_3 = Me$, Et, -CH₂- R_3O $R_$

Scheme 50.

51). Subsequently, Kim et al. reported a regioselective synthesis of 1-arylnaphthalene derivatives (**70**) from N-tosylaziridines (**69**), which, in turn, were obtained by reacting the appropriate N-tosylimine with cinnamyl bromides obtained from the Baylis—Hillman adducts. The pathway for the final product (**70**) was mediated via aziridine ring opening followed by a Friedel—Crafts-type reaction and elimination of p-toluene-sulfonic acid (Scheme 52).

An unusual formation of 2-amino-2,3-dihydrobenzofurans (71) from an appropriate substrate has been reported by Kim et al. (Scheme 53). The required substrate was prepared by reacting ethyl nitroacetate with the appropriate allyl bromide and in the presence of DABCO via an acid-promoted (TFA and H_2SO_4 mix) oxygen-atom-transfer process.⁷⁹

More recently, Lee et al. described the synthesis of 2-(9-fluorenyl)acrylic acid derivatives (**72**) via an intramolecular Friedel—Crafts reaction of the Morita—Baylis—Hillman adducts of 2-biphenylcarbaldehydes, as shown in Scheme 54.⁸⁰

Scheme 51.

Scheme 52.

OH
$$CO_2R'$$
 ArH , $60-70$ °C, $3h$ CO_2Et ArH , $60-70$ °C, $3h$ $Ar = Ph, 2,5-Me2C6H3, 2,5-(MeO)2C6H3 71$

Scheme 53.

$$\begin{array}{c} X \\ \\ HC \equiv C-COY \\ Bu_4NI, ZrCI_4 \text{ or } TiCI_4, Me_2S \\ \text{or } TiBr_4, Me_2S \text{ or } \\ CHO \\ \hline \\ CHO \\ \hline \\ CHO \\ \hline \\ CHO \\ \hline \\ CH_2 = CHCO_2Me \\ DABCO, (HOCH_2CH_2)_3N \\ 80-84\% \\ \hline \\ X = H, CI, OMe; Y = OMe, OEt, NMe_2; Z = I, CI, B \\ \end{array}$$

Scheme 54.

Das et al. reported a one-pot synthesis of allyl sulfides (73) from the acetates of the Baylis—Hillman adducts by treatment with benzenethiol in the presence of a catalytic amount of 15% NaOH and TBAI in DMSO. The base-mediated hydrolysis of the ester group followed by TFAA-promoted cyclization yielded (*Z*)-3-(4-methoxybenzylidene)thiochroman-4-one (74). Compound 74 was found to display promising antifungal activity (Scheme 55).⁸¹

Kanakam et al. demonstrated the preparation of methylenedinaphthyl bis-chromanones (75) via reaction between methylene dinaphthol and allyl bromide followed by saponification and TFA-mediated cyclization. These products showed efficient luminescence in the blue light region and elicited significant activity against *Staphylococcus aureus* and *Streptococcus faecalis* (Scheme 56).⁸²

8. Cycloaddition reactions

The presence of a double bond in the Baylis—Hillman adduct makes it a suitable substrate for the cycloaddition reactions. Initially, investigations were limited to 1,3-dipolar cycloadditions of nitrile oxides, 83 but, during the last few years, applications of the Baylis—Hillman derivatives for cycloaddition reactions have grown tremendously.

8.1. 1,3-Dipolar cycloadditions

Fisera et al. reported that the 1,3-dipolar cycloaddition of a *C*-phenyl-*N*-methyl nitrone to the Baylis—Hillman adducts proceeded with complete regioselectivity to afford the corresponding diastereomeric 3,5,5-trisubstituted isoxazolidines

Scheme 55.

(76) in good yields (Scheme 57). 84 They observed that the use of microwave irradiation helped in increasing the rate of the reaction. 85 Later, Das et al. successfully synthesized the isoxazolines (77) in high yields via the reaction of the Baylis—Hillman adducts with aldoximes in the presence of CAN (Scheme 58). 86

Raghunathan et al. have reported the synthesis of novel spiroheterocycles (**78**, **79**, and **80**) via a 1,3-dipolar cycloaddition reaction of nonstabilized azomethine ylides generated by the decarboxylative condensation of di- and tri-ketones with sarcosine and the Baylis—Hillman adducts. The reaction sequence, which is described in Scheme 59, was performed in refluxing toluene and proceeded with high regioselectivity and stereoselectivity. Interestingly, when the same reaction was performed in refluxing methanol, an unusual nucleophilic attack of the hydroxy group on a ketone carbonyl occurred, resulting in the products **81** and **82** (Scheme 60).⁸⁷ Very recently, the same

workers have reported the results of similar studies conducted with the Baylis—Hillman adducts of ninhydrin with sarcosine and proline. Since the yields were low in the conventional method of refluxing in methanol, they investigated the use of microwave conditions. They have reported that the time of the reaction can be drastically reduced by performing the reaction on Mont. K-10 with a significant improvement in yields and with high regioselectivity. The synthesis of a number of spiropyrrolidine/pyrrolizidine derivatives (83–86) developed by these workers is shown in Scheme 60.

Sreedhar et al. have reported a Cu(I)-catalyzed one-pot regioselective synthesis of 1,4-disubstituted 1,2,3-triazoles (87) in high yields. ⁸⁹ Their methodology involved nucleophilic displacement of the acetyl group in the Baylis—Hillman acetate with sodium azide followed by 1,3-dipolar cycloaddition of terminal alkynes in PEG (Scheme 61). Simultaneously, Chandrasekhar et al. demonstrated the synthesis of similar

Scheme 57.

OH EWG + R'
$$\frac{1}{1!}$$
 R' $\frac{1}{71-81\%}$ R = H, 2-Cl, 2-NO₂; R' = 2-Cl, 3-NO₂, 4-OMe $\frac{1}{71-81\%}$ R $\frac{1}{1!}$ R'

Scheme 58.

Scheme 59.

Scheme 60.

Scheme 61.

triazoles (88) by using Cu(0) and CuSO₄ in ethanol (Scheme 61). 90

Lee et al. have elegantly utilized the Baylis—Hillman derivatives for performing intramolecular 1,3-dipolar cycloadditions. ⁹¹ They have successfully synthesized 6-carbomethoxy-1,2,3,5-tetrahydroimidazo[2,3-*b*][1,3]benzodiazocines (**89**) in good yields by the intramolecular 1,3-dipolar cycloaddition of

2-(1-aziridinylmethyl)-3-(2-ureidophenyl) acrylic acid methyl ester, as delineated in Scheme 62. Simultaneously, they accomplished the synthesis of 5*H*-1,2,3-triazolo[4,3-*a*][2]benzazepines (**90**) in good yields via an intramolecular 1,3-dipolar cycloaddition of the allylic azido group with the alkyne group present on the aromatic ring (Scheme 63). ⁹² Alternatively, they generated the alkyne moiety at the allylic position from the

CHO
$$NO_{2} = \frac{\text{AZIIIdine, Et}_{3}N}{\text{THF, rt, 30 min}} = \frac{\text{SW Pd/C, H}_{2}}{\text{SW}} = \frac{\text{EtOH, rt, 2h}}{\text{63\%}} = \frac{\text{SW Pd/C, H}_{2}}{\text{EtOH, rt, 2h}} = \frac{\text{CO}_{2}\text{Me}}{\text{63\%}} = \frac{\text{FW Pd/C, H}_{2}}{\text{63\%}} = \frac{\text{CO}_{2}\text{Me}}{\text{FW Pd/C, H}_{2}} = \frac{\text{FW Pd/C, H}_{2}}{\text{63\%}} = \frac{\text{FW Pd/C, H}_{2}}$$

Scheme 63.

Baylis—Hillman acetates via a Grignard reaction and utilized it for intramolecular cyclization with the azido group, which was present on the phenyl ring, leading to the generation of 5-carbomethoxy-4H-1,2,3-triazolo[1,5-a][1]benzazepines (91) in low yields, as shown in Scheme 64. ⁹³ In a continuation of their studies, they have recently reported the synthesis of 4-carbomethoxy-naphtho[2,1-c]isoxazoles (92) via an intramolecular 1,3-dipolar cycloaddition of 2-alkynylphenylallyl nitrones (Scheme 65). ⁹⁴

A novel regioselective synthesis of a variety of functionalized 3-spiropyrrolidine oxindoles (93 and 95) and 3-spiropyrrolizidines (96 and 97) from the Baylis—Hillman adducts of

isatin and heteroaldehydes via a [3+2] cycloaddition of azomethine ylides was reported by Shanmugam et al. (Schemes 66 and 67). 95

Bakthadoss et al. utilized the allyl bromides afforded from the Baylis—Hillman adducts for a highly regio- and stereoselective synthesis of tricyclic chromeno[4,3-b]pyrrolidine (98 and 99) frameworks via a substitution reaction followed by the in situ formation of an imine, decarboxylation, and [3+2] cycloaddition sequence, as outlined in Scheme 68. 96

The synthesis of a series of chromene[4,3-b]pyrroles (100) through an intramolecular 1,3-dipolar cycloaddition reaction of an azomethine ylide with the dipolarophile afforded from

Scheme 64.

OAC
$$CO_2Me$$
 $\frac{NaNO_2}{rt, 1.5-10 \text{ h}}$ CO_2Me $\frac{PhNCO, Et_3N}{PhMe, rt, 1 \text{ h}}$ CO_2Me $\frac{PhNCO, Et_3N}{NO_2}$ $\frac{$

Scheme 65.

Scheme 66.

Scheme 67.

$$\begin{array}{c} \text{Ar} & \text{CO}_2\text{Me} \\ \text{Br} & \text{HO} \\ \end{array} \begin{array}{c} \text{K}_2\text{CO}_3, \text{MeCN}, \\ \text{rt, 1 h} \\ \hline \\ 79\text{-}97\% \\ \end{array} \\ \text{Ar} & \text{CO}_2\text{Me} \\ \text{CHO} \\ \end{array} \begin{array}{c} \text{MeNHCH}_2\text{CO}_2\text{H}, \\ \text{MeCN, reflux, 5 h} \\ \hline \\ 65\text{-}92\% \\ \end{array} \\ \text{MeO}_2\text{C} \\ \end{array} \\ \text{98} \\ \text{Ar} & \text{Ph, 4-Me-C}_6\text{H}_4, 4\text{-Et-C}_6\text{H}_4, 4\text{-i-Pr-C}_6\text{H}_4, 4\text{-F-C}_6\text{H}_4, 2\text{-Cl-C}_6\text{H}_4, 3\text{-Cl-C}_6\text{H}_4} \\ \text{Ar} & \text{CN} \\ \end{array} \\ \text{Pr} & \text{OHC} \\ \text{CN} \\ \text{HO} \\ \end{array} \begin{array}{c} \text{K}_2\text{CO}_3, \text{MeCN}, \\ \text{rt, 1 h} \\ \hline \text{81-95\%} \\ \end{array} \\ \text{OHC} \\ \text{Ar} & \text{NC} \\ \end{array} \\ \text{OHC} \\ \text{Ar} & \text{NC} \\ \end{array} \\ \text{OHC} \\$$

Scheme 68.

OH
$$Ar = Ph, 4-Cl-C_6H_4, 4-OMe-C_6H_4; EWG = CO_2Me, CN$$
 $R = H, Br, OMe$ $R = Ph, Br, OMe$

Scheme 69.

the Baylis—Hillman adducts, as described in Scheme 69, was accomplished by Ramesh and Raghunathan. They observed that the use of ultrasonic radiation for this particular reaction increases the efficiency and the products were generated in better yields.

8.2. Diels-Alder reactions

The Diels-Alder reaction of the Baylis-Hillman adducts with cyclopentadiene was first reported by Aggarwal et al.⁹⁸ They observed that the thermal conditions favored the exo isomer (101) over the endo isomer (102) and the ratio was controlled by a balance of steric and electronic factors. The use of a Lewis acid had no effect on this ratio, although, in the presence of EtAlCl₂, the ratio of the exo to the endo form was reversed. Interestingly, they reported that the Diels-Alder reaction of the Baylis-Hillman adducts with isoprene under thermal conditions gave a mixture of stereo- and regio-isomer, whereas the use of EtAlCl₂ furnished essentially a single diastereoisomer and regioisomer (103) in all cases. Initially, as compared to ethyl and cyclohexyl, the yield of 103 bearing phenyl substitution was moderate, but the use of 2,6-di-tertbutylpyridine in the reaction as base effectively increased the yield to quantitative (Scheme 70).

Later, Nair and Abhilash reported the synthesis of polycyclic aromatic hydrocarbons from the hemiacetal derived from the Baylis—Hillman reaction of *o*-phthaldehyde.⁹⁹ Strategically, the synthesis of the naphthalene (**104**) was achieved by DMAD-catalyzed dehydration of the hemiacetal leading to the isobenzofurans, which were trapped with several electron-deficient dienophiles (Scheme 71).

Lu et al. have reported novel approaches involving the cycloaddition of electron-deficient olefins with electron-deficient allylic compound as the three-carbon unit via a phosphorus ylide. Initially, they observed that the phosphonium bromide salt derived from the allyl bromide generated from the Baylis-Hillman adducts of formaldehyde reacts with N-phenylsuccinimide to yield the annulated cyclopentene (105) via (3+2) a cycloaddition (Scheme 72). 100 Consequently, they demonstrated their strategy to be successful with several substrates. A variety of allyl bromides, acetates, and O-Boc-protected derivatives of Baylis-Hillman adducts were treated with different electron-deficient alkenes to yield several cyclopentenes including the spiro derivatives. Subsequently, Lu et al. extended the scope of their methodology to the convenient construction of bridged nine-membered carbacycles (106). 101 The synthesis of the (3+6) cycloadducts was achieved by the reaction of allylic compounds with tropone in the presence of Ph₃P and K_2CO_3 as additives (Scheme 73). More recently, the reactions of the same allylic compounds with 2-substituted-1,1-dicyanoalkenes for the synthesis of highly substituted cyclopentenes (107) have also been reported by these workers (Scheme 74). 102 They found that their strategy was successful only with dicyanoalkenes, since replacement of the cyano group led to unsatisfactory results. Additionally, although regioselective

Scheme 72.

Scheme 70.

Scheme 71.

X = Br, CI, OAc, OBoc; EWG = CO_2Me/Et , COR **106** R = Ph, 4-Me-C₆H₄, 4-Cl-C₆H₄, 4-Br-C₆H₄, 4-OMe-C₆H₄

Scheme 73.

$$X = Br. OAc. OBoc$$
 $X = Br. OAc. OBoc$
 $X = Br. OAc. OBoc$

R = 2-naphthyl, 4-OMe $-C_6H_4$,

Scheme 74.

 $4-NO_2-C_6H_4$, $2,4-Cl_2-C_6H_3$, 2-furyl, n-Pr

and stereoselective formation of cyclopentene was successful with Ph₃P, better yields and regioselectivity were achieved by using EtPh₂P.

Blechert et al. described the synthesis of cis-fused carbobicyclic compounds (108) via an intramolecular (4+2) cyloaddition reaction of compounds 109, which were, in turn, obtained from the corresponding Baylis—Hillman derivatives via enyne cross-metathesis with terminal alkynes, as outlined in Scheme 75. 103

Lee et al. described the synthesis of divinylquinolinediones (110) by subjecting dihaloquinolinediones to a Baylis—Hillman-type reaction. They demonstrated that these divinylquinolinediones (110) undergo a thermal 6π electrocyclization to yield the benzo[g]quinolines (111) (Scheme 76).

Jorgensen et al. demonstrated the utility of the allyl amine (112) for the Diels—Alder reaction with 2,3-dimethyl butadiene for the synthesis of cyclohexene derivative (113) with high diastereoselectivity (Scheme 77).

8.3. (n+n) Annulations

A one-pot, regioselective synthesis of polysubstituted phenols (114) from the acetates of the Baylis—Hillman adducts and dimethyl 1,3-acetonedicarboxylate was described by Kim et al. via a formal [3+3] annulation strategy (Scheme 78). One Substituted phenol derivatives such as (115) were also synthesized from nitroalkane-substituted Baylis—Hillman derivatives of acrylates via a [4+2] annulation strategy, in which the nitroalkane derivative served as the four-carbon unit and the Michael acceptor (alkyl vinyl ketone or alkyl acrylate) as a two-carbon unit (Scheme 79). The nitroethane-substituted Baylis—Hillman derivatives of methyl vinyl ketone

$$\begin{array}{c} OX \\ & \downarrow \\ N \end{array} = \begin{array}{c} OX \\ & \downarrow \\ N \end{array} + \begin{array}{c} EWG \\ & \downarrow \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\$$

Scheme 75.

Scheme 76.

R = Ph, $4-NO_2-C_6H_4$, 2-Pyridyl, 2-naphthyl, Et, i-Pr; EWG = $CO_2Met-Bu$, CN

Scheme 77.

Scheme 78.

Scheme 79.

or ethyl vinyl ketone were demonstrated to be excellent precursors for the regioselective synthesis of polysubstituted benzenes (**116** and **117**), as shown in Schemes 80 and 81, respectively. Recently, Kim et al. have successfully synthesized polysubstituted nitrobenzene derivatives (**118**) from the Baylis—Hillman acetates via a [3+3] annulation strategy, in which 1,3-dinitroalkanes served as the 1,3-dinucleophilic component, while the Baylis—Hillman acetates featured as the 1,3-dielectrophilic unit, as shown in Scheme 82.

Shi et al. reported an abnormal aza-Baylis—Hillman reaction between *N*-tosylimines and allenes. They observed that the DABCO-catalyzed Baylis—Hillman reaction of ethyl 2,3-butadienoate with *N*-tosylimines yielded the azetidine derivatives (119) through a [2+2] cycloaddition process. On changing the base to DMAP, however, the reaction proceeded quickly to furnish dihydropyridines (120) via a [4+2] cycloaddition reaction (Scheme 83). Later, Shi and Zhao studied the aza-Baylis—Hillman reaction of *N*-(arylmethylene)-diphenylphosphinamides with various activated alkenes and observed that the Bu₃P-promoted Baylis—Hillman reaction with phenyl acrylate furnished 5-methylene-6-oxo-2-aryl-

piperidines (**121**) along with normal Baylis—Hillman adducts, whereas a similar reaction with phenyl vinyl ketone yielded highly substituted cyclopentenes (**122**), as shown in Scheme 84. ¹¹² Subsequently, a detailed study of the aza-Baylis—Hillman reactions of *N*-tosylated aldimines with activated allenes and alkynes in the presence of various Lewis base promoters was carried out by these workers. ¹¹³ The PPhMe₂-mediated aza-Baylis—Hillman reaction afforded 2,5-dihydro-1*H*-pyrroles (**123**), as delineated in Scheme 85.

The aza-Baylis—Hillman reaction between *N*-tosylaldimine and 3-methylpenta-3,4-dien-2-one was also described by Shi and Zhao. Interestingly, although the DMAP-catalyzed reaction afforded normal Baylis—Hillman adducts, the Bu₃P-mediated Baylis—Hillman reaction yielded 1,2,3,6-tetrahydropyridines (**124** and **125**), as shown in Scheme 86.

Kim et al. reported a two-step synthesis of polysubstituted benzenes in moderate yields starting from the Baylis—Hillman derivatives (126) obtained via an S_N2' reaction of ethyl cyanoacetate. The first step involved the base-promoted reaction of methyl/ethyl vinyl ketone to obtain a cyclohexene derivative (127), which, upon treatment with DBU in DMF, yielded

Scheme 80.

OAC R
$$R_1CH_2NO_2$$
 Ph R_1 NO_2 R_2 R_3 R_2 R_3 R_4 R_5 $R_$

Scheme 81.

OAC R
$$K_2CO_3$$
, DMF NO_2 K_2CO_3 , DMF NO_2 NO_2

Scheme 82.

 $Ar = Ph, 4-Me-C_6H_4, 4-F-C_6H_4, 4-Cl-C_6H_4, 4-Br-C_6H_4, 3-F-C_6H_4, 3-NO_2-C_6H_4, 2,3-Cl_2-C_6H_3, 1-naphthylogeness and the second substitution of t$

 $Ar = 4 - Et - C_6H_4, \ 4 - OMe - C_6H_4, \ 4 - F - C_6H_4, \ 4 - Cl - C_6H_4, \ 4 - Br - C_6H_4, \ 4 - NO_2 - C_6H_4, \ PhCH = CH_4 - CH_4$

Scheme 84.

Scheme 85.

$$\begin{array}{c} \text{DMAP, DMSO} \\ \text{ArCH=NTs} + \\ \text{O} \end{array} \\ \begin{array}{c} \text{Bu}_{3}\text{P, DCE} \\ \text{80 °C, 30 min} \\ \text{Ar} \end{array} \\ \text{Ar} = \text{Ph, 4-Me-C}_{6}\text{H}_{4}, \text{4-Et-C}_{6}\text{H}_{4}, \text{4-F-C}_{6}\text{H}_{4}, \text{4-Br-C}_{6}\text{H}_{4}, \text{2-naphthyl} \end{array} \\ \begin{array}{c} \text{Ts} \\ \text{Ar} \\ \text{Ar} \end{array} \\ \begin{array}{c} \text{Ts} \\ \text{Ar} \\ \text{Ar} \\ \text{O} \end{array} \\ \begin{array}{c} \text{O} \\ \text{O} \end{array} \\ \begin{array}{c} \text{O}$$

Scheme 86

Scheme 87.

the highly substituted benzene (128), as shown in Scheme 87.115

9. Radical cyclizations

The radical cyclization has been extensively applied to the Baylis—Hillman chemistry for the generation of heterocyclic or carbocyclic derivatives. Shanmugam et al. have demonstrated the use of radical cyclization for the synthesis of several oxygen-containing heterocycles. Most of these syntheses have been accomplished from the suitable propargyl derivatives obtained by the reaction of propargyl alcohol with the Baylis—Hillman adduct on Mont. K-10. Interestingly, they reported the isomerization of the propargyl compounds if the reaction on Mont. K-10 was pursued for longer periods (Fig. 8). Radical cyclization of the alkenyl propargyl ether with freshly distilled *n*-Bu₃SnH in the presence of AIBN

yielded the vinylstannane through 5 *exo-trig* cyclization. The vinylstannane on protiodestannylation in the presence of 1 N HCl afforded the *exo*-methylene-tetrahydrofurans (**129** and **130**) (Scheme 88). They extended this strategy for the synthesis of pyrans (**131**) and oxepenes (**132**), as shown in Scheme 89. Very recently they have demonstrated the scope of this strategy for the synthesis of 3-heteroaryl-methyl-substituted tetrahydrofurans **133** (Scheme 90). 119

Subsequently, in a slight modification of this strategy in which the same propargyl derivatives were obtained via a reaction in the presence of catalytic H_2SO_4 in place of Mont. K-10, Kim et al. described the synthesis of similar *exo*-methylenetetrahydrofurans. Further, they utilized these compounds to accomplish the synthesis of 3,4-disubstituted 2,5-dihydrofurans (134) via saponification, and halolactonization followed by final spontaneous decarboxylation and β , β -disubstituted- α -methylene- γ -butyrolactones (135) via PCC/Ac₂O-mediated

Figure 8. Isomerization of propargyl derivatives at higher temperatures.

Scheme 88

Scheme 89.

Scheme 90

oxidation, as depicted in Scheme 91.¹²⁰ They have also disclosed the regioisomeric synthesis of two types of methylene-tetrahydropyrans (**136** and **137**) in high yields by reacting propargyl derivatives and allyltributylstannane followed by the strategy of vinyl radical cyclization, as shown in the reaction sequence in Scheme 92.¹²¹

In an alternative strategy to afford propargyl derivatives for the radical cyclization reaction, Kim et al. treated the S_N2' reaction products with propargyl bromide in the presence of NaH. These propargyl derivatives (138) in the presence of $n\text{-Bu}_3\text{SnH}$ underwent radical cyclization to furnish the *exo*methylene-cyclopentanes (139). They also attempted a similar reaction with propargyl derivatives (140) generated from the S_N2 reaction products, but this failed to yield the desired

compounds (Scheme 93). ¹²² Recently, they have reported a successful synthesis of dihydronaphthalene (**141**) scaffolds using these cyclopentanes (**142**) as the starting substrates. Treating **142** with I_2 or NBS leads to the addition of halogen on the methylene group, which, on radical cyclization in the presence of allyltributylstannane yielded the dihydronaphthalenes (**141**), as shown in Scheme 94. Their attempt to carry out the cyclization in the presence of n-Bu₃SnH was unsuccessful. ¹²³

In another variation of the strategy, Kim et al. have successfully transformed the 2-bromophenol-substituted Baylis—Hillman derivatives into 3,3-disubstituted-2,3-dihydrobenzofurans (143) in good yield (Scheme 95). Later, they reported an interesting synthesis of hexahydrofuro[2,3-*b*] furans (144) and

OH propargyl alcohol Ar
$$CO_2Me$$
 $\frac{H_2SO_4, CH_2CI_2}{rt, 6 \text{ h}, 54-75\%}$ Ar $= \text{Ph}, 4\text{-Cl-C}_6H_4$ Ar CO_2Me $\frac{H_2SO_4, CH_2CI_2}{rt, 6 \text{ h}, 54-75\%}$ Ar CO_2Me $\frac{1. Bu_3SnH, AlBN, neat, 80 °C, 20 min}{2. dil. HCl, ether, rt, 20 min, 62-87\%}$ $\frac{MeO_2C}{4r}$ $\frac{PCC, Ac_2O}{55-77\%}$ $\frac{MeO_2C}{55-77\%}$ $\frac{FCC}{4r}$ $\frac{Ar}{4r}$ $\frac{Ar}{4r}$

Scheme 91.

Scheme 92.

Scheme 93.

OAC Ar = Ph, 4-Me-C₆H₄ EWG = CO₂Me/Et, CN
$$\begin{array}{c} Ar \\ EWG_2 \\ EWG_1 \\ \hline \\ Ar = Ph, 4-Me-C_6H_4 \\ \hline \\ Ar = Ph, 4-Me-C_6H_4$$

Scheme 94.

OAc Ph EWG + Br
$$\frac{R}{HO}$$
 R $\frac{K_2CO_3, THF}{63-89\%}$ Ph EWG R $\frac{Bu_3SnH, AlBN}{62-79\%}$ PhH, reflux $\frac{R}{62-79\%}$ EWG = $\frac{CO_2Me}{E}$, CN; R = H, Cl, Me

Scheme 95.

OAC
$$Ph$$
 CO₂Me CO_2 Me CO

Scheme 96.

hexahydrofuro[2,3-b]pyrans (145) from the bromoacetals in the presence of n-Bu₃SnH via the Ueno-Stork reaction (Scheme 96). 125

Our research group was the first to report the radical cyclization involving the halide of the aromatic ring and the methylene group of the Baylis—Hillman adducts during the successful synthesis of a new isoxazolo-benzazulene system (146) from the derivatives of 3-(2-bromophenyl)-4-

isoxazolecarbaldehydes (Scheme 97). We found that this reaction did not take place with the corresponding chloro derivatives. Later, Chattopadhyay et al., using a similar strategy, reported the synthesis of enantiopure benzo-fused nine-membered oxacycles (147 and 148). Initially, they transformed the *O*-2-bromobenzylated-1,2:5,6-di-*O*-isopropylidene glucofuranoside and its analogs into noraldehydes, which, upon Baylis—Hillman reaction, followed by successful radical

Scheme 97.

cyclization, yielded the macrocycles (Scheme 98). 127 Very recently, they reported n-Bu₃SnH-mediated transformation of the enamides generated from the Baylis—Hillman adducts of 2-bromobenzaldehydes into dihydropyrido[2,1-a]isoindolones (149), as shown in Scheme 99. 128

Atom-transfer radical polymerization was employed in the cyclopolymerizaton of a symmetrical dimethacrylate (150) under the influence of CuBr/PMDETA by Acar et al. (Scheme 100).¹²⁹

The radical cyclization has been utilized for the generation of isochroman (152) and tetrahydroisoquinoline derivatives (154) from the Baylis—Hillman acetates by Kim et al. The S_N2' reaction of 2-bromobenzyl alcohols with the acetates yielded 151, which were transformed into the isochroman (152) in the presence of $n\text{-Bu}_3\text{SnH}$, as shown in Scheme 101.

A similar reaction sequence with 2-bromobenzyl tosylamides (153) resulted in the formation of the tetrahydroisoquinolines (154) in good yields. ¹³⁰

In another strategy, the derivatives afforded via the S_N2' reaction of tosylamine with the Baylis—Hillman acetates followed by alkylation with 1,4-dibromo-2-butane were cyclized in the presence of n-Bu₃SnH to afford the N-tosyl-3,3,4-trisubstituted-pyrrolidines (155) by this group (Scheme 102). Furthermore, they also employed a similar protocol for the generation of tetrahydrofuran systems (156). Accordingly, the Baylis—Hillman adducts were reacted with cis-1,4-butanediol followed by bromination and radical cyclization resulting in 156, albeit in low yields (Scheme 103). 131

10. Heck coupling reactions

The Heck coupling reaction, another carbon—carbon bond-forming strategy, involving the coupling of halo-substituted arenes with alkenes or alkynes has been successfully applied for the generation of cyclic systems from the Baylis—Hillman derivatives. Lee et al. described the synthesis of indanones (157 and 158) from the Baylis—Hillman adducts of 2-iodo-

Scheme 98.

$$R = H, 5-F; R' = Me, Et$$

$$CO_{2}R'$$

$$R = H, 5-F; R' = Me, Et$$

$$CO_{2}R'$$

$$R = H, 5-F; R' = Me, Et$$

$$CO_{2}R'$$

$$R = H, 5-F; R' = Me, Et$$

$$CO_{2}R'$$

$$R = H, 5-F; R' = Me, Et$$

$$CO_{2}R'$$

$$R = H, 5-F; R' = Me, Et$$

$$R = H, 5-F; R$$

Scheme 99

Scheme 100.

Scheme 101.

Scheme 102.

OH Ph
$$CO_2Me$$
 + OH Mont K-10 OH Mont K-10 OH Mont K-10 OH Ph, reflux, 4 h Ph OH Ph, reflux, 4 h Ph OH 156 5%

Scheme 103.

benzaldehydes via intramolecular Heck cyclization in the presence of Pd(OAc)₂ and (*o*-Tol)₃P, whereas, under similar reaction conditions the Baylis—Hillman adducts of 2-cyclohexen-1-ones afforded the 1-hydroxyfluorenes (**159**) (Scheme 104). ¹³³ The successful transformation of aza-Baylis—Hillman adducts of 2-halosulfonamides into highly constrained bicyclic 6,7-dihydro-5-thia-6-aza-benzocycloheptene 5,5-dioxides (**160**)

via the intramolecular Heck reaction was reported by Vasudevan et al. (Scheme 105). Alternatively, these workers utilized similar products originating from 2-halo-benzaldehydes to synthesize substituted indenes (**161**) in moderate-to-good yields (Scheme 106). The construction of 1*H*-inden-1-ones (**162**) was also successfully described by She et al. via a sequential intramolecular Heck reaction followed by an aerial

Scheme 104.

Scheme 105.

$$R = H, 4-CF_{3}$$

$$R = 4-Me-C_{6}H_{4}; 4-F-C_{6}H_{4}$$

$$R = Ar S O Ar S NH O Pd(Ph_{3})_{4}$$

$$R = 4-Me-C_{6}H_{4}; 4-F-C_{6}H_{4}$$

$$R = Ar S O NH O Pd(Ph_{3})_{4}$$

$$Et_{3}N, THF$$

$$S O Pd(Ph_{3})_{4}$$

$$S O Pd(Ph_{3})_{4}$$

$$Et_{3}N, THF$$

$$S O Pd(Ph_{3})_{4}$$

$$Et_{3}N, THF$$

$$S O Pd(Ph_{3})_{4}$$

$$S$$

Scheme 106.

oxidation of allylic alcohols of Baylis—Hillman adducts of 2-halo-benzaldehydes, as shown in Scheme 107. 135

Kim et al. have earlier described the synthesis of **149** from the enamides. ¹²⁸ In an extension of this work, they have reported a successful transformation of these enamides into novel pentacyclic benzoazepino[2,1-*a*]isoindole derivatives (**163** and **164**) via a double carbopalladation using Hecktype cyclization conditions, as shown in Scheme 108. ¹³⁶

Very recently, these workers have also reported the synthesis of 2-arylquinolines **166** via a Pd-mediated sequential Heck-type cyclization and concomitant aerobic oxidation from the S_N2 reaction product **165** provided by the reaction between

Scheme 107.

the Baylis—Hillman acetates and 2-bromoanilines, as shown in Scheme 109.¹³⁷ With the aim of the extending the scope of their strategy, they also demonstrated the synthesis of tetrahydropyridines (**168**) following a similar reaction sequence. The amides (**167**) afforded by the reaction of the Baylis—Hillman acetates with tosylamide followed by alkylation with 2,3-dibromopropene or allyl bromide and subsequent Pd-catalyzed cyclization resulted in tetrahdropyridines (**168** and **169**) in moderate yields.

11. Ring-closing metathesis reactions

Ring-closing metathesis in Baylis—Hillman chemistry was introduced by Paquette and Mendez-Andino for the synthesis of α -methylene- γ -lactones fused to medium and large rings. ¹³⁸ Kim et al. have published a series of papers on the application of RCM on a variety of Baylis—Hillman derivatives to generate several heterocyclic scaffolds. Initially, they subjected the *O*- and *N*-allylic derivatives to RCM in the presence of Grubbs' second-generation catalyst to synthesize

Scheme 108.

OAc
$$Ar = Ph, 4-Me-C_6H_4, EWG = CO_2Me/Et$$
 $Ar = Ph, 4-Me-C_6H_4, EWG = CO_2Me/Et$ $Ar = 2-Br-C_6H_4$ $Ar = 2-Br-C_6H_4$ $Ar = Ph, 4-Me-C_6H_4$ $Ar = Ph, 4-Me-C_6H_4$ $Ar = 2-Br-C_6H_4$ $Ar = Ph, 4-Me-C_6H_4$ $Ar = 2-Br-C_6H_4$ $Ar = 2-Br-C_6H_4$ $Ar = Ph, 4-Me-C_6H_4$ $Ar = 2-Br-C_6H_4$ $Ar = 2-Br-C_6H_4$ $Ar = 2-Br-C_6H_4$ $Ar = Ph, 4-Me-C_6H_4$ $Ar = 2-Br-C_6H_4$ $Ar = 2-Br-C_6H_4$

Scheme 109.

Scheme 110.

dihydrofurans (170) or dihydropyrroles (171) (Scheme 110).¹³⁹ Further, they generated other allyl derivatives (172), which served as useful synthons for the RCM reaction to yield cyclopentenes (173) in excellent yields (Scheme 111).¹⁴⁰ Recently, the same workers have demonstrated the utility of RCM for the generation of furo[3,4-*c*]pyran (174) and pyrano[3,4-*c*]pyrrole (175) rings, as shown in Scheme 112.¹⁴¹

Balan and Adolfsson reported the RCM of allyl amino derivatives (176) originating from the aza-Baylis—Hillman products via Grubbs' second-generation catalyst under microwave conditions to afford N-tosyldihydropyrrole (177) (Scheme 113). Subsequently, Lamaty et al. demonstrated the synthesis of similar products, but these contained a 2-SES group instead of a tosyl moiety. They showed that, in the presence of t-BuOK, these dihydropyrroles can be easily

deprotected and aromatized to generate pyrroles (178) (Scheme 114). Later, in another report, they compared the SES group with a tosyl group for the preparation of nitrogen-containing five-membered rings obtained by the aza-Baylis—Hillman/alkylation/RCM route. They observed that deprotection of tosyl-protected pyrrolines gave only pyrroles, whereas deprotection of similar SES-protected derivatives furnished either pyrroles or pyrrolines depending on the deprotecting conditions. Indeed, they also hydrogenated the SES-protected pyrrolines to yield pyrrolidines (179) with excellent diastereoselectivity (Scheme 115). Later

Krafft et al. described a new and efficient approach to functionalized hetero- and carbocyclic alkenoles, which are also afforded via an intramolecular Baylis—Hillman reaction. They generated the required synthon via a quinuclidine-promoted

Scheme 111.

OH
$$Ar = Ph, 4-Me-C_6H_4$$
 $Ar = Ph, 4-Me-C_6H_4$ $Ar = Ph, 4-Me-C_6$

Scheme 112.

$$\begin{array}{c} O \\ Ar \\ H \\ TsNH_2 \end{array} \begin{array}{c} 3\text{-HQD, Ti(OPr)_4} \\ MS \ 4 \ A \ , \text{t-PrOH, rt} \\ 95\text{-}99\% \end{array} \begin{array}{c} \text{NHTs} \\ Ar \\ \end{array} \begin{array}{c} CO_2\text{Me} \\ \hline K_2\text{CO}_3, \text{DMF} \\ 90\text{-}99\% \end{array} \begin{array}{c} TsHN \\ Ar \\ \end{array} \begin{array}{c} CO_2\text{Me} \\ \hline K_2\text{CO}_3, \text{DMF} \\ 90\text{-}99\% \end{array} \begin{array}{c} TsHN \\ Ar \\ \end{array} \begin{array}{c} CO_2\text{Me} \\ \hline Ar = Ph, 3\text{-CI-C}_6\text{H}_4, 4\text{-OMe-C}_6\text{H}_4, 2\text{-Pyridyl} \\ \hline 4\text{-NO}_2\text{-C}_6\text{H}_4, 2\text{-naphthyl} \end{array}$$

Scheme 113.

Scheme 114.

Scheme 115.

Morita—Baylis—Hillman reaction of alkenyl aldehydes, which, on RCM reaction under the influence of Grubbs' second-generation catalyst, yielded the cycloalkenoles (180) in excellent yields (Scheme 116). They reported that their methodology was successful for the generation of eight- or nine-membered rings. More recently, Wang et al. have reported a novel, simple, and eco-friendly method to provide cyanonaphthalenes (181) via a sequential Claisen rearrangement, Baylis—Hillman reaction, and RCM in the presence of Grubbs' second-generation catalyst (Scheme 117). 146

The sugar-based Baylis—Hillman adducts were efficiently utilized to prepare a library of diverse α,β -unsaturated λ -lactones (182) by Krishna and Narsingam using the RCM reaction (Scheme 118). 147

Trost and Brennan successfully employed the dienes generated for the Morita—Baylis—Hillman adducts in a Pd-catalyzed asymmetric alkylation with Meldrum acid to generate a product, which, on further alkylation, furnished a bisalkylated derivative **183** in moderate yields over two steps, as shown in Scheme 119. The product **183** upon treatment with a second-generation Grubbs' catalyst afforded the 1,2-disubstituted alkene **184**. 148

Recently, Kim et al. disclosed an expeditious route for the synthesis of pyrrolidines and tetrahydrofurans (185 and 186), starting from the suitably modified Baylis—Hillman adducts involving RCM followed by hydrogenation or radical cyclization and subsequent hydrogenation protocols, as shown in Scheme 120. 149

Scheme 116.

Scheme 117.

$$H_3CH_2CO \longrightarrow R \xrightarrow{\text{LiAlH}_4, \text{AlCl}_3, \\ \text{ether}, 0 \, ^{\circ}\text{C}, 2 \text{ h} \\ \text{63-68\%} \longrightarrow HO \longrightarrow R \xrightarrow{\text{DIEA}, \text{CH}_2\text{Cl}_2} \xrightarrow{\text{O} \, ^{\circ}\text{C} \text{ to rt}, 10 \text{ h}} \xrightarrow{\text{S1-75\%}} OH \xrightarrow{\text{CH}_2\text{Cl}_2, 36 \text{ h}} OH \xrightarrow{\text{CH}_2\text{Cl}_2, 10 \text{ h}} OH \xrightarrow{\text{CH}_$$

Scheme 118.

Scheme 119.

Scheme 120.

Doddi and Vankar described efficient syntheses of two pyrrolidine-based imino sugars (**187** and **188**) initiating from the Baylis—Hillman adduct of (*R*)-2,3-*O*-isopropylideneglyceraldehydes. The key steps included the regiospecific ami-

nation (via an S_N 2 reaction), RCM and diastereospecific dihydroxylations, as delineated in Schemes 121 and 122. These azasugars were reported to be moderate inhibitors of glycosidase. ¹⁵⁰

Scheme 121.

Scheme 122.

$$\begin{array}{c} \text{MeO}_{2}\text{CO}_{2}\text{Me} \\ \text{R} & \text{CO}_{2}\text{Me} \\ \text{R} & \text{S}_{8}\text{-91}\% \\ \text{R} & \text{Me} \text{N}_{3}\text{-Me}\text{-}\text{C}_{6}\text{H}_{4}, 3\text{-CF}_{3}\text{-}\text{C}_{6}\text{H}_{4}, 4\text{-NO}_{2}\text{-}\text{C}_{6}\text{H}_{4} \\ \text{R} & \text{N}_{4}\text{-NO}_{2}\text{-}\text{C}_{6}\text{-}\text{H}_{4} \\ \text{N}_{5}\text{-Ne}\text{-}\text{C}_{6}\text{-}\text{Ne}\text{-}\text{C}_{6}\text{-}\text{Ne}\text{$$

Scheme 123.

Scheme 124.

Donohoe et al. reported the transformation of the Baylis—Hillman adducts into mixed acetates, which, upon RCM reaction followed by aromatization, furnished the 2,3-substituted furans (**189**) in excellent yields, as shown in Scheme 123. ¹⁵¹

Several other research groups have also employed the RCM strategy for the synthesis of a variety of natural products from the Baylis—Hillman derivatives. Theses approaches are discussed in Section 21.

12. Epoxide formation

Baylis—Hillman derivatives have been successfully exploited for the generation of several substituted oxirane compounds. A highly diastereoselective epoxidation of allylic diols to provide the *trans*-epoxides (190) from the Baylis—Hillman adducts was reported by Coelho et al. The high

$$\begin{array}{c} \text{NS} \\ \text{HN} \\ \text{Ph} \end{array} \begin{array}{c} \text{CO}_2 \text{Me} \\ \hline \\ \text{NS} = 4\text{-NO}_2 \text{C}_6 \text{H}_4 \text{SO}_3 \end{array} \begin{array}{c} \text{HN} \\ \text{NS} \\ \text{Ph} \\ \text{NS} \end{array} \begin{array}{c} \text{CO}_2 \text{Me} \\ \text{Ph} \\ \text{NS} \end{array}$$

Scheme 125.

$$H_2N$$
 H_2N H_2N

Scheme 126.

PhI=O, cat. KBr
$$H_2O$$
, rt, 5-8 h $R=NO_2$, CF₃, CI, Me; EWG = CO₂Me, CN $R=NO_3$

Scheme 127.

anti-diastereoselectivity was attributed to the formation of an intramolecular hydrogen bond (Scheme 124).¹⁵² Raheem and Jacobson described the formation of a *syn*-epoxide (**191**) from the aza-Baylis—Hillman adduct under the influence of *t*-BuOOH and Triton B, as shown in Scheme 125.¹⁵³

Metzner et al. developed a highly diastereoselective organocatalytic synthesis of functionalized vinyl epoxides (**192**), displaying a Morita—Baylis—Hillman backbone by means of sulfonium ylide epoxidation of aldehydes from the easily available α-(bromomethyl)acrylamide derivatives (Scheme 126). 154

Das et al. described the synthesis of acyloxiranes (193) from the Baylis—Hillman adducts in the presence of iodosobenzene, which was utilized as an oxidizing agent for the two-fold oxidation of a secondary alcohol followed by epoxidation of the generated enone (Scheme 127). ¹⁵⁵ A catalytic amount of KBr was required for the success of this reaction, since iodosobenzene alone did not give products.

13. Aldol condensation

The products originating from the S_N2 or S_N2' reaction of activated methylene compounds with the Baylis—Hillman acetates have been successfully employed for the aldol condensation to afford the cyclic scaffolds. Amri et al. reported the first aldol condensation reaction in the Baylis—Hillman chemistry to provide an easy access to 4-alkylidene-2-cyclohexen-1-ones. The first also reported that the S_N2' reaction of the Baylis—Hillman acetates of acrylates with 2,4-pentane-dione yielded a product, which, under the influence of LiHDMS, furnished the 4-arylidenecyclohexane-1,3-diones (194) (Scheme 128). Alternatively the products afforded from the S_N2' reaction between the Baylis—Hillman acetates of methyl vinyl ketone and diethyl malonate were transformed into the same products via a similar protocol. Changing the nucleophile to ethyl cycloalkanone-2-carboxylate in the S_N2'

Scheme 129.

OAC O DABCO, acetyl acetone Ar = Ph, 4-Cl-C₆H₄; R = Me, Et
$$\begin{array}{c} OAC O \\ Ar \end{array}$$

$$\begin{array}{c} OAC O \\ Ar \end{array}$$

Scheme 130.

reaction of Baylis—Hillman acetates of alkyl vinyl ketones yielded 3-alkylidenebicyclo[3.2.1]octan-8-ones (**195**) albeit in slightly lower yields and longer reaction times (Scheme 129). 158

Subsequently, Kim et al. demonstrated that the S_N2 reaction of Baylis—Hillman acetates of alkyl vinyl ketones with acetylacetone in the presence of DABCO led to a synthon, which, on refluxing with K_2CO_3 in ethanol, gave 4-methylene-2-cyclohexenones (196) (Scheme 130). Further treatment of these products with I_2 in methanol led to the formation of the benzene derivatives (197). Recently, these workers have developed the synthesis of 3,4,5-trisubstituted pyridines (198) from the Baylis—Hillman acetates involving an aldol reaction as the key step (Scheme 131). 160

Zhu et al. reported a double aza-Baylis—Hillman reaction between per- (or poly-) fluorophenyl aromatic aldimines

with methyl vinyl ketone and found that the double aza-Baylis—Hillman adduct (199) was a viable precursor for the construction of fluorine-containing 4-alkylidene-2-cyclohexen-1-ones (200 and 201) under mild conditions and with good yields (Scheme 132).¹⁶¹

Recently, Kim et al. have demonstrated the synthesis of 3-benzylidene-cyclohexenes (202) from the Baylis—Hillman derivatives via a Michael addition of the appropriately substituted Baylis—Hillman derivatives to the activated alkenes followed by a DBU-promoted intramolecular aldol reaction (Scheme 133). 162

14. Claisen rearrangement

Basavaiah's group demonstrated the synthesis of alkylidine dialkanoates from the reaction between Baylis-Hillman

Scheme 131.

$$R = 2-Br, \ 2-Cl; \ Ar_F = C_6F_5, \ 4-Cl-C_6F_4$$

$$R = 2-Br, \ 2-Cl; \ Ar_F = C_6F_5, \ 4-Cl-C_6F_4$$

$$DABCO \\ THF \\ 65-87\%$$

$$R = \frac{Ar_FHN}{(1 - 1)^{10}} O \xrightarrow{K_2CO_3, EtOH} R \xrightarrow{Ar_FHN} O \xrightarrow{Ar_FHN} R \xrightarrow{Ar_FHN} O \xrightarrow{Ar_FHN}$$

Scheme 132.

OH O 1. acetylation 2.
$$CH_2(CO_2R)_2$$
 Ph CO_2R_2 DBU/MeCN reflux, 1-20 h CO_2R_2 EWG = COMe/Et, CO_2Me/Et , CN Ph CO_2R_2 202

Scheme 133.

adduct and triethyl orthoacetate as the first example of a Claisen rearrangement in the Baylis-Hillman chemistry. 163 Recently, Kim et al. reported a PPA-promoted Claisen rearrangement for the synthesis of 3-benzylidene-3,4-dihydro-1*H*-quinolin-2one (203), 3-benzylquinolin-2-ol (204), 4-amino-2-benzylideneindan-1-one (205), and 1-amino-9a,10-dihydro-4bH-indeno-[1,2-a]inden-9-one (206) skeletons from the Baylis-Hillman derivatives of acrylates. 164 They observed that a slight difference in the electron density of the aniline moiety causes strikingly different reaction pathways, as methyl- or methoxysubstituted anilines favor a Friedel-Crafts reaction, whereas chlorine-substituted or unsubstituted anilines cyclize via an intramolecular cyclization between the amino and the ester group (Scheme 134). Subsequently, the same workers reported a synthesis of quinolines (207) from the Baylis-Hillman derivatives of methyl or ethyl vinyl ketone via a sequential aza-Claisen rearrangement, intramolecular cyclization and DBU-promoted isomerization (Scheme 135). 165

With an objective of our research group being to develop an easy and scaleable approach to the starting material for the anti-tubercular compound R207910, we have also developed

a strategy to obtain 3-arylidene-2-quinolones (208). In our effort to replace PPA (difficult to handle on a large scale), we discovered that TFA is a better choice to effect the Claisen reaction. In contrast to the PPA-promoted reaction, the TFA-mediated cyclization was not affected by the change in the substituent-dependent electron density of the phenyl ring (Scheme 136). ¹⁶⁶ This led to the development of a high-yielding and column chromatography-free synthesis of 2-methoxy-3-arylquinolines. Interestingly, a similar synthetic protocol with the Baylis—Hillman derivatives of acrylonitrile provided the 3-aryl-2-amino quinolines (209) via a tandem Claisen rearrangement, intramolecular cyclization, and subsequent isomerization.

15. Lactonization reaction

Over the years, several Baylis—Hillman derivatives have been utilized elegantly for the generation of the α -methylene- γ -butyrolactone, ring, which is ubiquitously present in several alkaloids and terpenoids. ¹⁶⁷ Kim et al. have successfully developed the synthesis of α -methylene- γ -butyrolactones

Scheme 134.

OAc 1. DABCO NHAr 1. PPA, CICH
$$_2$$
CH $_2$ Cl 80-90 °C, 2 h

R = Me, Et
Ar = Ph, 4-Me-C $_6$ H $_4$, 4-OMe-C $_6$ H $_4$, 2,3-Me $_2$ -C $_6$ H $_3$, 1-naphthyl

Scheme 135.

Scheme 137.

(210) via the reaction between allyl bromides and a variety of reactive carbonyl compounds under the influence of In and NH₄Cl, as shown in Scheme 137.¹⁶⁸ Indeed, with ninhydrin or *N*-benzylisatin, they demonstrated the formation of spiro analogs (211).

Subsequently, Kabalka et al. utilized the *syn*-homoallylic alcohols, prepared via a one-pot, cross-coupling/allylboration reaction on the Baylis—Hillman acetates, for the synthesis of *cis*- and *trans*- α -methylene- γ -lactones. Interestingly, using CBr₄/Ph₃P as the ring-closing agent stereoselectively furnished the trans-isomer (212), while the use of TsOH as the lactonization agent gave the cis-isomer (213) (Scheme 138). ¹⁶⁹

A library of mono- or disubstituted α -methylene- γ -lactones (214) was generated via a straightforward two-step reaction by Gouault et al. (Scheme 139). ¹⁷⁰ Their fluorous-phase synthetic strategy employed flourous acrylates as the starting materials and removal of fluorous alcohol via cycloelimination in the final step.

Very recently, another innovative strategy for the synthesis of α -methylene- γ -lactones with 99% selectivity has been accomplished by Ramachandran and Pratihar. The treatment of (*E*)-methyl or aryl 2-boramethyl-2-butenoate afforded

from the Baylis—Hillman adduct of cyclohexane-carbaldehyde under the influence of Yb(OTf)₃ or TFA gave the cis- β , γ -disubstituted- α -methylene- γ -butyrolactones (215), as shown in Scheme 140. A similar reaction in the presence of In(OTf)₃ or TfOH furnished the γ -substituted- α -E-alkylidene- α -methylene- γ -butyrolactones (216). Porto and Coelho described a simple and straightforward approach for the synthesis of an α -methylene- γ -butyrolactone (217) from the Baylis—Hillman adduct of 2,3-isopropylidine-D-glyceraldehyde, as depicted in Scheme 141. 172

Indeed, several other workers have also described excellent syntheses of arylidene-butyrolactones. The transformation of triple-bond tethered methyl cinnamates to arylidene lactones (218) under the influence of iodine via iodolactonization was reported by Kim et al. (Scheme 142). They also demonstrated the success of their strategy with NBS and H₂SO₄ as electrophiles to provide bromo- or proton-substituted enol lactones (219 and 220). Simultaneously, by sequential introduction of an appropriate vinyl moiety on the Baylis—Hillman acetates by Grignard reaction and treatment with benzene in the presence of H₂SO₄, these workers demonstrated the synthesis of dihydronaphthalenes (221) via the lactonization

Scheme 138.

Scheme 139.

Scheme 140.

Scheme 141.

Scheme 142.

OAC
$$R_{1}^{\parallel}$$
 R_{2}^{\parallel} R_{1}^{\parallel} R_{1}^{\parallel}

Scheme 143.

process, as delineated in Scheme $143.^{174}$ In order to provide evidence for the route of the reaction, they transformed the starting vinyl derivatives into the lactones (222) via an epoxy intermediate using m-CPBA.

More recently, Kim et al. described the transformation of a diene (223) afforded from Baylis—Hillman acetates into arylidene lactones (224) via an epoxy intermediate, as shown in Scheme 144. Moreover, these workers have also successfully achieved the synthesis of arylidene lactones (225) from

the *N*-tosylaziridines originating from the reaction between the allyl bromide and *N*-tosylamine (Scheme 145). This transformation was accomplished under the influence of LiClO₄ in acetonitrile. Interestingly, they reported that, if a similar reaction was performed in the presence of aniline, the *N*-tosylaziridine was transformed into the 3-arylidene-4,5-disubstituted- γ -butyrolactam derivatives (**226**) in excellent yields. Earlier, Maier et al. reported the synthesis of a spirolactone (**227**) from the vinylogous Baylis—Hillman product via

OAC
$$CO_2Me$$
 + $R = H$, Me , CI PPh_3 , K_2CO_3 $MeCN$, rt , 30 h $58-68\%$ Ph CO_2Me Ph CO_2Me

Scheme 144.

Ar = Ph, 4-Cl-C₆H₄, 2,4-Cl₂-C₆H₃

$$Ar = Ph, 4-Cl-C6H4, 2,4-Cl2-C6H3$$

$$Ar = Ph, 4-Cl - C6H4, 2,4-Cl2-C6H3$$

Scheme 145.

Scheme 146.

a chelation-controlled epoxide-opening strategy under the influence of Ti(O*i*-Pr)₄ (Scheme 146). 1777

Similar to five-membered lactones, there are various reports describing the synthesis of six-membered lactones. We have reported a mild and convenient synthesis of substituted α -methylene δ -valerolactones (229) from the substrates (228) originating by an S_N2 reaction of acetylacetone with the Baylis—Hillman acetates. These products undergo one-pot saponification of the ester, and reduction of the keto group followed by intramolecular ring closure in an aqueous medium, as outlined in Scheme 147. These lactones are viable

precursors for generating spiro-isoxazolines (**230**) via 1,3 dipolar cycloaddition of nitrile oxides on the double bond. ¹⁷⁹ We have successfully employed the diketo derivatives (**228a**) also for the synthesis of 3-methylene-3,4-dihydropyran-2-ones (**231**) via P_2O_5 -mediated cyclization (Scheme 148). ¹⁸⁰ Conversely, similar products (**228b**) obtained from methyl acetoacetate via a similar reaction sequence stereoselectively yielded *trans*-5-methylene-dihydropyran-2,6-diones (**232**). Kim et al. reported the successful lactonization of the S_N2' products of Baylis—Hillman acetates and deoxybenzoins to generate 3-arylidene-5,6-disubstituted-3,4-dihydropyran-2-ones (**233**), as

Ar = various aromatic and isoxazolecarbaldehydes

Scheme 147.

Scheme 148.

Scheme 149.

delineated in Scheme 149.¹⁸¹ These pyranones were oxidized in the presence of PCC to yield 3-aryl- α -pyrones (234). A similar protocol with α -tetralone led to the mono- and bis-products (Scheme 150). The bis-product was lactonized to yield the same lactone (235), which has been reported earlier by Basavaiah et al. ¹⁸² Simultaneous to this work, Kim et al. reported the formation of 7,8-dihydro-6*H*-chromene derivatives (236), starting from the dimedone as nucleophile, according to the reaction sequence shown in Scheme 151.¹⁸³

16. Reductive cyclization

The strategy of reductive cyclization has found extensive applications in the Baylis—Hillman chemistry with respect to the synthesis of a variety of nitrogen- and oxygen-containing heterocyclic systems. The syntheses of several nitrogen-containing heterocycles have been achieved by reduction of the nitro group followed by cyclization with an appropriate

functional group. This nitro group can be present either on the aromatic moiety generally at the *ortho* position to the formyl function on which the Baylis—Hillman reaction has been performed or on the side chain as a part of the nucleophile, which has been attached through an S_N2 or S_N2' reaction.

16.1. Reduction of aromatic nitro groups

The reductive cyclization involving an aromatic nitro group in the Baylis—Hillman chemistry has evolved extensively. O'Dell and Nicholas reported the formation of a mixture of *N*-formyl indolines (**237**) and indoles (**238**) via reduction of the nitro group by using 53 atm of CO and [CpFe(CO)₂]₂ (Fp₂) as the catalyst. ¹⁸⁴ Later, however, these workers achieved the synthesis of 3-substituted quinolines (**239**) in good yields via chemoselective reduction of the nitro group followed by cyclization by employing 6 atm of CO and 10 mol % of Fp₂ (Scheme 152). ¹⁸⁵

Scheme 150.

OAC Ph
$$CO_2Me$$
 R_2 CO_3 , DMF Ph OH CO_2Me Ph OH OH Ph OH

Scheme 151.

Scheme 152.

Scheme 153.

Very recently, Kim et al. have reported that the Baylis—Hillman adducts of 2-nitrobenzaldehydes on treatment with SnCl₂ in 1,4-dioxane lead to a mixture of indoles (**240**) and benzisoxazolines (**241**) (Scheme 153). Earlier, they have reported that 2-quinolones (**242**) were formed when the same reaction was performed in alcohol. 187

Basavaiah et al. were the first to demonstrate that the Baylis—Hillman derivatives produced from 2-nitrobenzaldehyde are excellent precursors to functionalized 2-quinolones and quinolines. Subsequently, Kim et al. reported the formation of 3-substituted-4-hydroxyquinoline *N*-oxides (**243**) from similar substrates. Unlike Basavaiah et al., who have employed Fe/AcOH as reducing agent, these workers used TFA in the presence of triflic acid for this purpose (Scheme 154). Later, they reported that, in the presence of Zn–NH₄Cl, 3-substituted-quinoline *N*-oxides (**244**) were generated from these substrates in moderate yields (Scheme 154). Recently, Coelho et al. have reported their studies on the mechanistic details of the TFA/TfOH-mediated synthesis of quinoline *N*-oxides (**243**) by ESI-(+)-MS(/MS) monitoring.

Basavaiah et al. demonstrated that the Baylis—Hillman adducts obtained from the reaction of 2-nitrobenzaldehyde with cycloalkenones could be easily transformed into different annulated quinoline systems (245 and 246) via reduction of the nitro group with Fe/AcOH (Scheme 155). They extended the scope of this strategy further by developing the synthesis of 3-benzoylquinoline derivatives (247) from the

Baylis—Hillman adducts of chromenones and 2-nitrobenzaldehyde, as shown in Scheme 156. 193

Recently, Liu et al. have described a mild and efficient direct nucleophilic substitution of the Baylis—Hillman acetates derived from cyclic enones with indoles in the presence of AgOTf as a catalyst. The reaction provided highly α -regioselective indole derivatives (248). The reaction products of the 4-nitro-substituted indole derivatives were utilized for the synthesis of novel azepino[4,3,2-cd]indoles (249) via a one-pot nitro-group reduction followed by an in situ aza-Michael addition reaction, as delineated in Scheme 157. 194

More recently, Kaye et al. have also reported, the Pd/C-mediated reductive cyclization of Baylis—Hillman adducts of 2-nitrobenzaldehyde (250). They observed that the chemo- and regioselectivity of the cyclization is influenced by the choice of both the substrate and the reagent systems (Scheme 158). ¹⁹⁵

We have recently reported an alternative route to quinolines (251 and 252) from the S_N2 reaction products of Baylis—Hillman acetates of substituted 2-nitrobenzaldehydes and carbonyl-group-containing nucleophiles, as shown in Scheme 159. We have observed that treatment of these products with $SnCl_2$ triggers a tandem reaction, in which the reduction of the nitro group is followed by a regioselective intramolecular cyclization and subsequent dehydration. Interestingly, this study indicated that the generated amino group has preference for the activated carbonyl for cyclization in the order: R=Me>Ph>O-alkyl.

Scheme 154.

Scheme 155.

Scheme 156.

Scheme 157.

R₁ = H, OMe; R₂ = H, OMe, NO₂, -OCH₂O-, R₃ = H, Cl OH, OMe, R₄ = H, Cl, NO₂; EWG = COMe/Et, CO₂Me/Et, CN, SO₂Ph

Scheme 158.

$$\begin{array}{c} \text{OAc} \\ \text{PWG} \\ \text{EWG} \\ \text{PMO}_2 \\ \text{THF/H}_2\text{O, rt, 1 h} \\ \text{78-85\%} \\ \text{R} = \text{H, 5CI, 3,4-OCH}_2\text{O}; \text{EWG} = CO}_2\text{Me, CN} \\ \text{228a or 228b} \\ \end{array} \begin{array}{c} \text{SnCl}_2, \text{MeOH} \\ \text{reflux, 1h} \\ \text{R}_1 = \text{Me/Ph} \\ \text{R}_2 = \text{Ph/OEt} \\ \text{50-86\%} \\ \text{SnCl}_2, \text{MeOH} \\ \text{reflux, 1 h} \\ \text{SnCl}_2, \text{MeOH} \\ \text{reflux, 1 h} \\ \text{R}_1 = \text{R}_2 = \text{Me/Ph} \\ \text{SnCl}_2, \text{MeOH} \\ \text{reflux, 1 h} \\ \text{R}_1 = \text{R}_2 = \text{Me/Ph} \\ \text{SnCl}_2, \text{MeOH} \\ \text{reflux, 1 h} \\ \text{R}_1 = \text{R}_2 = \text{Me/Ph} \\ \text{SnCl}_2, \text{MeOH} \\ \text{reflux, 1 h} \\ \text{SnCl}_2, \text{MeOH} \\ \text{reflux, 1 h} \\ \text{NO}_2 = \text{NO}_2 + \text{NO$$

Scheme 159.

Scheme 160.

Very recently, Basavaiah et al. have demonstrated an elegant synthesis of an azocine moiety via a one-pot, multi-step protocol. Initially, an S_N2' reaction between 1,3-cycloalkane-diones and Baylis—Hillman adducts of 2-nitrobenzaldehyde, followed by reduction with Fe/AcOH and subsequent cyclization, yielded the azocine derivatives (253) in good yields (Scheme 160). 197

16.2. Reduction of aliphatic nitro groups

Reductive cyclization with aliphatic nitro groups in the Baylis—Hillman derivatives has also been exemplified in variety of procedures. Basavaiah and Rao first reported a one-pot facile synthesis of γ -lactams (254) from the product obtained via an S_N2' reaction of Baylis—Hillman acetates and

nitroalkanes. ¹⁹⁸ The Fe/AcOH-mediated intramolecular reductive cyclization was accomplished by involving the amine generated from the nitro group and the ester of the side chain, as depicted in Scheme 161.

Subsequently, analogous to this strategy, Kim et al. used similar substrates afforded by the Baylis—Hillman adducts of methyl vinyl ketone for the synthesis of 4-benzylidene-2,5-dimethyl-3,4-dihydro-2*H*-pyrroles (255), as shown in Scheme 162. ¹⁹⁹ Further, they transformed these pyrroles (255) into bicylic lactams (256) in moderate yields. Later, they extended this strategy and successfully synthesized 2-benzylidene-7a-alkyl-tetrahydropyrrolizine-3,5-dione derivatives (257) from the Michael adduct afforded by the reaction between nitroalkanoate and acrylate. However, the similar analog obtained via reaction of nitroalkanoate with methyl vinyl

OAC O
$$rt$$
, 12 h rt , 13 h rt

 $R = Ph, 4-Me-C_6H_4, 4-Et-C_6H_4, 4-Cl-C_6H_4, 4-OMe-C_6H_4, 2-Cl-C_6H_4, 1-naphthyl, Pr, Hex$

Scheme 161.

Scheme 162.

Scheme 163.

ketone upon reduction of the nitro group to amine furnished the dihydropyrrole (258) and its N-oxide derivative (259) (Scheme 163).

We have recently reported a facile synthesis of 3-exo-methylene pyrrolidinones (260) via In/HCl-mediated reductive cyclization of Baylis—Hillman derivatives afforded from the S_N2 reaction of Baylis—Hillman acetates with ethyl nitroacetate, as shown in Scheme 164. We discovered that, in these substrates, the secondary nitro group undergoes a partial reduction to the oximes (261) in the presence of $SnCl_2 \cdot 2H_2O$. These oximes were transformed into pyrrole derivatives (262) in low yields. Interestingly, similar nitro derivatives originating

from the Baylis—Hillman acetates of substituted 2-nitrobenz-aldehydes led to the formation of 3-vinylindoles (263) under the influence of $SnCl_2 \cdot 2H_2O$, due to chemoselective reduction of the aromatic nitro group to amine and the aliphatic nitro group to oxime.

More recently, Kim et al. have also transformed the nitro derivatives obtained by an $S_{\rm N}2$ reaction between nitroalkanes and allyl bromides into 3-methylene-pyrrolidinones (264) via Fe/AcOH-mediated reductive cyclization (Scheme 165).²⁰²

In an extension to our work, we have recently reported the synthesis of benzazepines (265 and 266) by $SnCl_2 \cdot 2H_2O$ -promoted reductive cyclization of the S_N2' reaction product of ethyl

Scheme 164.

Scheme 165.

Scheme 166.

nitroacetate and nitroethane with the Baylis—Hillman acetates of substituted 2-nitrobenzaldehydes, as shown in Scheme 166.²⁰³ We observed that the formation of the 1- or 3-benzazepines was influenced by the substituent on the phenyl ring in the products obtained as derivatives of ethyl nitroacetate, but such an influence was absent in the products derived from nitroalkanes. Surprisingly, it was observed that the 1*H*-1-benzazepines were unstable in light or on silica gel and rearranged to the isoquinolines (267) (Scheme 166). This unusual transformation proceeded by a plausible mechanism, as shown in Figure 9.

16.2.1. Reduction of other functional groups

Besides reductive cyclizations involving the nitro group, there are several other reports of generating cyclic systems via this strategy in Baylis—Hillman chemistry. The syntheses of substituted hydrindanones (268) and decalones (269) via Li/NH₃-promoted intramolecular cyclization of the Baylis—Hillman derivatives, originating from 4-pentenal and cycloal-kenones in moderate yields were described by Marko et al. (Scheme 167).²⁰⁴ They proposed that the formation of the products proceeded via a radical reaction.

Figure 9.

Scheme 167.

Scheme 168.

OH
$$EtO_2C$$
 CN H_2 , Raney-Ni EtO_2C NH $MeOH, 3 h$ Ar CO_2Me $Ar = Ph, 4-Me-C6H4, 2-Cl-C6H4, 4-Cl-C6H4, 4-F-C6H4, 4-F-C6H4,$

Scheme 169.

Shanmugam et al. demonstrated the transformation of bromo derivatives of Baylis—Hillman adducts of isatin into functionalized diastereomeric 3-spirocyclopropane-2-indolones (**270**) via NaBH₄-mediated reductive cyclization (Scheme 168). 205

We have reported a facile synthesis of 6-oxo-4-aryl-piper-idine-3-carboxylates (271) from the S_N2 products of ethyl cyanoacetate with the Baylis—Hillman acetates under Pd—C-mediated hydrogenation. These substrates were demonstrated to be excellent precursors for the synthesis of a new bicyclic heterocyclic system, namely 5-methyl-4-oxo-6-aryl-3-azabicyclo[3.1.0]hexane-1-carboxylates (272) (Scheme 169).

17. Hydrolysis of nitrile groups

The hydrolysis of the nitrile group, present in the Baylis—Hillman adducts derived from acrylonitrile, or as part of the nucleophile added on to the Baylis—Hillman derivatives, has been utilized in different ways to obtain heterocycles, especially containing an amide linkage.

Kim et al. have utilized this strategy for the generation of several heterocycles from the Baylis—Hillman derivatives obtained from the nucleophilic addition of nitrile-containing nucleophiles. The H₂SO₄-mediated hydrolysis of the nitriles of 3-aryl-2-cyano-methyl 1,2-propenoates provided the succinimide derivatives (273) via amide formation.²⁰⁷ Similar succinimides (273) were reported by us via TFA/H₂SO₄-pro-

moted hydrolysis followed by cyclization with NaH (Scheme 170). 208

Kim et al. also described the synthesis of glutarimide derivatives (274) from the cinnamyl bromides according to the reaction shown in Scheme $171.^{207}$ An S_N2 reaction of malononitrile on the acetate of the Baylis–Hillman adduct of acrylate followed by hydrolysis afforded a diamides, which was cyclized by these workers to yield the arylideneglutarimide (275) (Scheme 172).

We have also utilized the hydrolysis of the nitrile group in the Baylis—Hillman derivatives as a strategy for generating glutarimide derivatives. Interestingly, the hydrolysis with TFA/H₂SO₄ yielded 4,5-disubstituted-3-methylene-piperidine-2,6-diones (276), while hydrolysis during the FeCl₃-promoted reaction provided 4-substituted-3-methylene-piperidine-2,6-diones (277) (Scheme 173).

More recently, our group has achieved the synthesis of 3-methylene-2-pyridones (278) from the S_N2 reaction products of Baylis—Hillman acetates and acetylacetone or methyl acetoacetate via TFA/ H_2SO_4 -promoted hydrolysis of the nitrile group, as shown in Scheme 174. Peplacing the nucleophile with ethyl cyclopentanone-2-carboxylate provided the annulated 2-pyridones (279) by the application of the same reaction protocol. These pyridones were demonstrated to be the precursors for synthesizing novel 2-pyridone-spiroisoxazoline derivatives in highly regio- and diastereoselective fashion.

Scheme 170.

Scheme 171.

Scheme 172.

Scheme 173.

Scheme 174.

18. Heterocyclization of substituted 3-aminopropanols and allyl amines

The nucleophilic addition of amines on the Baylis—Hillman adduct results in substituted 3-aminopropanols, while a similar addition on the Baylis—Hillman acetates or allyl bromides afforded substituted allyl amines. Both these substrates have been efficiently utilized for the synthesis of several nitrogen-containing heterocycles.

Hatakeyama et al. reported β -ICD-catalyzed asymmetric Baylis—Hillman reactions of aromatic imines with HFIPA and the successful transformation of the products to generate chiral β -lactams (280), as outlined in Scheme 175. 210

Kim et al. described the synthesis of 4-arylidene-2-substituted-isoxazolidin-5-ones (281) from *N*-hydroxyallyl amines via LiClO₄-promoted intramolecular cyclization involving hydroxy and ester groups (Scheme 176).²¹¹

Our group has described earlier a solid-phase methodology for the facile synthesis of annulated pyrimidinone derivatives (282) from the Baylis-Hillman acetate generated on Wang resin via reaction with diamine followed by treatment with cyanogen bromide and subsequent base-promoted cyclative cleavage (Scheme 177). 212 This strategy was simulated on automation and its scope was extended in a series of papers from our group. In this effort, we successfully utilized the Baylis-Hillman adducts (283, 288) of acrylonitrile and in most of the cases the different primary amines were added in an S_N2 or S_N2' fashion followed by LiAlH₄ or Raney-Ni-mediated reduction of the nitrile group to yield the diamines (such as 286 and 289), which on subsequent treatment with cyanogen bromide, yielded the tetrahydro-pyrimidines 284²¹³ and substituted imidazo[1,2-a]pyrimidin-2-ones (287 and 290) 214 according to the reaction sequences depicted in Schemes 178 and 179, respectively. Compounds 284 were easily transformed into the substituted imidazo[1,2-a]pyrimidin-2-ones 285 by treatment with ethylbromoacetate.

Following a similar reaction protocol, our research group has increased the scope of this strategy by the preparation of

$$\begin{array}{c} R \\ Ar \\ H \\ HFIPA \end{array} \begin{array}{c} GF_3 \\ GF_3 \\ GF_3 \\ HFIPA \end{array} \begin{array}{c} GF_3 \\ GF_3$$

Scheme 175.

$$\begin{array}{c} \text{OAc} \\ \text{R} & \text{CO}_2\text{Me} \end{array} \\ \begin{array}{c} \text{R'NHOH.HCI} \\ \text{DMF, Et}_3\text{N} \\ \\ \text{0-10 °C, 1-3 h} \\ \\ \text{51-88\%} \end{array} \\ \text{R} = \text{Ph, 4-Cl-C}_6\text{H}_4, 2-\text{furyl, 2-naphthyl, penl} \end{array} \\ \begin{array}{c} \text{R'} & \text{CO}_2\text{Me} \\ \text{NOH} \\ \text{reflux, 7-13 h} \\ \text{S3-100\%} \\ \text{R'} = \text{Me, Cy, Bn} \end{array} \\ \begin{array}{c} \text{R'} \\ \text{281} \end{array}$$

Scheme 176.

O OH AcCI, Py, O OAc 1,n-diaminoalkane DMF, rt, 15 h n = 2-4

16 examples
$$Ar = 2-NO_2-C_6H_4$$
, $3-NO_2-C_6H_4$, $2-furyI$, various isoxazoles $H_2N \not\downarrow I_n$

CNBr, DMF-EtOH rt, 12-30 h

 $H_2N \not\downarrow I_n$
 I_1
 I_2
 I_2
 I_1
 I_2
 I_2
 I_1
 I_2
 I_1
 I_2
 I_1
 I_2
 I_1
 I_2
 I_2
 I_1
 I_2
 I_1
 I_2
 I_2
 I_1
 I_2
 I_2
 I_1
 I_2
 I_1
 I_2
 I_1
 I_2
 I_2
 I_1
 I_1
 I_2
 I_1
 I_2
 I_1
 I_2
 I_2
 I_1
 I_1
 I_2
 I_1
 I_2
 I_1
 I_2
 I_2
 I_2
 I_1
 I_2
 I_1
 I_2
 I_1
 I_2
 I_1
 I_2
 I_1
 I_2
 $I_$

Scheme 177.

Scheme 178.

Scheme 179.

Scheme 180.

hexahydro-pyrimido[1,2-a]pyrimidin-2-ones (**292**) from the products (**291**) originating from the reaction of derivatized primary allyl amines with the Baylis—Hillman adduct obtained by reacting acrylonitrile and acetyl derivatives of the Baylis—Hillman adducts of acrylates (Scheme 180). These compounds exhibited significant antiparasitic activity.²¹⁵

The substituted 3-aminopropanols 293 derived from the reaction of primary amines with the Baylis—Hillman adduct of acrylonitrile on treatment with isocyanate furnished the urea derivatives 294. In the presence of a base, compounds 294 cyclized into the pyrimidinones 295, in which the benzyl moiety was lost during the process of cyclization. This loss was attributed to the presence of the hydroxyl group, since it was prevented on converting the hydroxyl group into a methoxy moiety. On the contrary, the substituted primary allyl amines 296 derived from the Baylis—Hillman acetates were easily transformed into the 4-iminium-pyrimidones 298 via treatment

with isocyanate and ring closure of the product (297), as shown in Scheme 181.²¹⁶ Some of these compounds showed excellent antibacterial activity.

More recently, substituted 3-aminopropanols 299 have been demonstrated to be excellent precursors for the generation of 5-benzyluracil derivatives 300. 217 The cyanamides afforded by compounds 299 undergo a base-promoted cascade reaction in which the nitrile is hydrolyzed due to the migration of hydroxyl group followed by intramolecular cyclization, as shown in Scheme 182. Protecting the hydroxyl group, however, as in compound 301, provides a stable cyanamide 302 and prevents the cascade cyclization. Treatment of 302 with hydroxylamine hydrochloride furnished compound 303. Alternatively, the allyl amine derivatives 304 yielded by the Baylis—Hillman acetates were regioselectively oximated at the 2-position via sequential treatment with cyanogen bromide and hydroxylamine hydrochloride to furnish the uracil derivatives 305.

Scheme 181.

Scheme 182.

More recently, we have demonstrated that the primary allyl amines **306** derived from the Baylis—Hillman acetates can be easily formylated with neat formamide to provide the *N*-formamides **307**. This reaction was utilized for the development of a one-pot procedure for the synthesis of 4-amino-5-methylpyrimidines (**308**). Interestingly, the product originating from

2-nitrobenzaldehyde was converted into the pyrimido[4,5-*b*]-quinoline (**309**), as shown in Scheme 183.²¹⁸

The preparation of 3-aryl-3-hydroxypyrrolidin-2-ones **311** and tricyclic 2-benzyl-9b-hydroxy-3,3a,5,9b-tetrahydro-2*H*-pyrrolo[3,4-*c*]quinoline-1,4-diones (**312**) starting from the Baylis—Hillman adducts (**310**) of isatin was successfully

formamide, 120 °C, 1 h then HCO₂NH₄, 140 °C, 2.5 h
Ar
$$CO_2$$
Me formamide, CO_2 Me formamide, CO_2 Me CO_2 Me

Scheme 183.

Scheme 184.

demonstrated by Kim et al. via the sequence shown in Scheme 184.²¹⁹

Kim and coworkers successfully developed the synthesis of 3,4-disubstituted pyridines (313) from the tosylamides of the Baylis—Hillman adducts of alkyl vinyl ketones.²²⁰ A Schweizer reaction of the tosylamine derivatives with vinyl-triphenylphosphonium bromide, elimination of tosyl group and the final 1,3-proton shift yielded the product in very good yields, as shown in Scheme 185. In a more recent report these workers disclosed another synthesis of polysubstituted pyridines (314) via a 3+2+1 annulation strategy as delineated in Scheme 186.²²¹ Although this synthesis did not involved allyl amines as substrate directly, it was an

extension of their previously described work (vide infra Scheme 131).

Very recently, these workers have reported a successful synthesis of pyrrole derivatives (316 and 318) from the substrates (315 and 317) afforded via an S_N2' and S_N2 reaction of the tosylamines with the Baylis—Hillman acetates. The procedure involved N-alkylation, Michael addition—elimination of p-toluenesulfonic acid, and finally DBU-mediated oxidative aromatization via the reaction sequence shown in Scheme 187.

The allyl amine derivative originating from the reaction of 1,2-phenylenediamine with the Baylis—Hillman acetate was utilized by Kim et al. to synthesize 3-(benzylidene)-1,5-benzo-diazepin-2-one. Our group has also demonstrated the

OH O 1. acetylation Ar R
$$\stackrel{\bigcirc}{=}$$
 PPh₃ $\stackrel{\bigcirc}{=}$ Ar $\stackrel{\bigcirc}{=}$ R $\stackrel{\bigcirc}{=}$ Cs₂CO₃, DMF $\stackrel{\bigcirc}{=}$ 120-130 °C, 2 h $\stackrel{\bigcirc}{=}$ Ar $\stackrel{\bigcirc}{=}$ R = Me, Et Ar = Ph, 4-Me-C₆H₄, 4-Cl-C₆H₄

Scheme 185.

R = Ph, Pent; R₁ = Me/Et; R₂ = Me/Ph; EWG = CO₂Me/Et, COMe, SO₂Me/Ph

Scheme 186.

Scheme 187.

 $Ar = Ph, 2-Cl-C_6H_4, 2-F-C_6H_4, 2,4-Cl_2-C_6H_3, 4-Me-C_6H_4, 4-Cl-C_6H_4, 4-Br-C_6H_4$

Scheme 189.

synthetic utility of derivatized allyl amines (**319** and **320**) from the Baylis—Hillman adducts for the generation of substituted 3-methylenebenzo[b][1,4]di-azepin-2-ones (**321** and **322**) and -benzo[b][1,4]di-azepine-2-ylamines (**323**) in moderate-to-good yields (Scheme 188). 224

Our research group has earlier reported that primary allyl amines originating from the reaction between Baylis—Hillman acetates of 3-aryl-5-formyl-isoxazole-4-carboxylates and benzylamine undergo an intramolecular cyclization to furnish a novel isoxazole-annulated system (324) (Scheme 189).²²⁵ The yield of the annulated system was, however, dependent on the nature of the EWG and was found to be highest with a *tert*-butoxy-carbonyl group.

Back et al. have described the Morita—Baylis—Hillman reaction of aldimines with 1-(*p*-toluenesulfonyl)-1,3-butadiene in the presence of 3-hydroxyquinuclidine and found that the *E*-isomers of the products (325) cyclize to yield the functionalized piperidines (326) (Scheme 190). ²²⁶ To improve the efficiency of the cyclization, a simultaneous equilibration of the

 $Ar = Ph, \ \, 4-CI-C_6H_4, \ \, 3-CI-C_6H_4, \ \, 2-CI-C_6H_4, \ \, 4-OMe-C_6H_4, \ \, 4-NO_2-C_6H_4, \ \, 4-CN-C_6H_4, \ \, 4-CN-C_6H_4, \ \, 4-NO_2-C_6H_4, \ \, 4$

Scheme 190.

(*E*)- and (*Z*)-isomers was effected by photo-isomerization. Later, Back et al. reported an aza-Morita—Baylis—Hillman reaction between substituted N-(phenylsulfonyl)aldimines and conjugated dienes activated by sulfone, ester, or ketone moieties to afford the highly functionalized allylic amine derivatives (327) (Scheme 191).²²⁷ Here, too, the *E*-isomer of the product from the dienyl sulfone and the dienoate underwent a facile intramolecular conjugate addition to produce the functionalized piperidines (328), whereas the adducts obtained from the dienones failed to cyclize under similar reaction conditions.

Muto et al. have synthesized a series of 6-substituted-4-sulfonyl-1,4-diazepane-2,5-diones (**329**) starting from the allyl amines, which, in turn, were generated from the corresponding Baylis—Hillman acetates via the introduction of an azide group followed by Ph₃P-mediated reduction (Scheme 192).²²⁸ These compounds exhibit good inhibitory activity against recombinant human chymase. Continuing with this work, Muto et al. also reported the synthesis of 4-aminocarbonyl-1,4-diazepane-2,5-diones (**330**), potent human chymase inhibitors, from the *N*-Boc protected allyl amine as shown in Scheme 193.²²⁹

Kim et al. reported the facile synthesis of 1,2,3,4-tetrasubstituted pyrroles (**331**) from substituted allyl amines afforded from the acetyl derivatives of Baylis—Hillman adducts via base-mediated cyclization followed by acid-catalyzed dehydration and concomitant double-bond isomerization, as shown in Scheme 194.²³⁰

Scheme 191.

Scheme 192.

 $R_1 = H$, Me; $R_2 = Bn$, CH_2Bn , $CH_2(Me)Ph$, $CH_2(Et)Ph$ $CH_2(i-Pr)Ph$, $CH_2(n-Pr)Ph$

 $Ar = Ph, 4-Me-C_6H_4, 4-Cl-C_6H_4$; $EWG = CO_2Me/Et$

Scheme 193.

OAC O Ph + H-N R
$$\frac{\text{K}_2\text{CO}_3, \text{ MeCN}}{\text{reflux, 1 h-7 d}}$$
 Ph OH $\frac{\text{TsOH, reflux, 10 h-2 d}}{\text{50-74}\%}$ EWG $\frac{\text{Fermion of the photons}}{\text{41-64}\%}$ R = CO₂Me/Et, Ph; EWG = CO₂Me/Et, COPh $\frac{\text{TsOH, reflux, 10 h-2 d}}{\text{N}}$ R $\frac{\text{TsOH, reflux, 10 h-2 d}}{\text{N}}$ EWG $\frac{\text{Solve of the photons}}{\text{N}}$ EWG $\frac{\text{So$

Scheme 194.

Scheme 195.

19. Ring transformations

We have reported the ring transformation of several Baylis—Hillman derivatives of substituted isoxazolecarbaldehydes via hydrogenation. Raney-Ni or Pd—C-promoted hydrogenolysis of the Baylis—Hillman adducts of 5- or 3-isoxazolecarbaldehyde leads to the enaminones **332** and **333**, respectively,

Scheme 196.

which were readily cyclized in the presence of a formic and acetic acid mixture to the 2,3-dihydrofurans (**334**) in excellent yields (Scheme 195).²³¹ In contrast, the hydrogenolysis of the Baylis—Hillman adducts of substituted 4-isoxazolecarbaldehydes provided the substituted 2-pyridinones (**335**), albeit in low yields (Scheme 196).

Subsequently, we were successful in transforming the enaminones afforded from the appropriate Baylis—Hillman derivatives of 3-isoxazolecarbaldehyde into the 2-pyrolidinones (336), 2-pyrolones (337), and pyrrolidines (338) in good yields, as depicted in Scheme 197.²³² Later, in another study, we have demonstrated the hydrogenolysis of the Baylis—Hillman derivatives of 3-(2-nitrophenyl)-5-isoxazolecarbaldehydes as a facile route for the synthesis of 2-substituted 4-aminoquinolines (339) (Scheme 198).²³³

Scheme 197.

NO₂ HO R Raney-Ni, MeOH
$$H_2$$
, rt, 30 psi, 5 h $R = H$, N-methyl piperazinyl H_2 R H_3 OH H_4 R H_4 R H_5 R H_5 R H_6 R H_7 R H_8 R H_9 R H_9

Scheme 198.

A DBU-promoted ring transformation of substituted 3-aminopropanols (**340**) of Baylis—Hillman adducts of 5-isoxazole-carbaldehydes to the pyrroles (**341**), via neighboring group participation, as shown in Scheme 199, was reported by our group.²³⁴

We have reported the transformation of enaminones derived via hydrogenolysis of Baylis—Hillman derivatives of isoxazolecarbaldehydes to the corresponding pyrazoles (352) in excellent yields by reacting them with hydrazine hydrate in refluxing conditions (Scheme 200).²³⁵

20. Miscellaneous reactions

Kamimura et al. described an $AgClO_4/MeI$ -mediated elegant stereoselective synthesis of β -lactams (343) from thiophenol-substituted Baylis—Hillman derivatives, as shown in Scheme 201. 236

Coelho et al. described an easy and straightforward alternative method of preparation of 3,4-substituted-isoquinolin-1(2H) ones (**344**) from the 2-bromo-substituted Baylis—Hillman derivatives, according to the reaction sequence shown in Scheme 202.²³⁷

Later, they described a palladium-mediated carbonylative cyclization reaction for the stereoselective synthesis of 3-alkenylphthalides (**345**) from the Baylis—Hillman adducts of *ortho*-halo-substituted benzaldehydes (Scheme 203).²³⁸ The quinoline-phthalide derivatives exhibited potent in vitro antiproliferative activity against human tumor cell lines.²³⁹

Scheme 199.

Ar
$$CO_2Me$$
 $Raney-Ni, H_2, Solid MeOH, 80 °C, 3 h RAP_2 RAP_2 RAP_2 RAP_2 RAP_2 RAP_2 RAP_2 RAP_2 RAP_3 RAP_4 RAP_3 RAP_4 RAP_4 RAP_4 RAP_5 RAP_4 RAP_5 RAP_4 RAP_5 RAP_4 RAP_5 RAP_5 RAP_5 RAP_6 $R$$

Scheme 200.

OAc O 1. cat. TsOH, PhMe, reflux, 2 h
$$2$$
 NHOBn $\frac{AgClO_4}{53-83\%}$ Ar = Ph, 4-Cl-C₆H₄, 1-naphthyl

Scheme 201.

$$\begin{array}{c} R_1 \\ R_2 \\ R_1 = R_2 = H, \text{ OCH}_2\text{O} \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_1 = R_2 = H, \text{ OCH}_2\text{O} \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_2 \\ \end{array} \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_1 \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_2 \\ \end{array} \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_1 \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_2 \\ \end{array} \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_1 \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_2 \\ \end{array} \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_2 \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_1 \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_2 \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ R_2 \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ \\ \text{OTIPS} \\ \\ \text{OTIPS} \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ \\ \text{OTIPS} \\ \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ \\ \text{OTIPS} \\ \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ \\ \text{OTIPS} \\ \\ \text{OTIPS} \\ \\ \text{OTIPS} \\ \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ \\ \text{OTIPS} \\ \\ \text{OTIPS} \\ \\ \text{OTIPS} \\ \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ \\ \text{OTIPS} \\ \\ \\ \text{OTIPS} \\ \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ \\ \\ \text{OTIPS} \\ \\ \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ \\ \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ \\ \\ \end{array} \\ \begin{array}{c} \text{OTIPS} \\ \\ \\ \end{array} \\$$

Scheme 202.

Scheme 203.

Kim et al. demonstrated the synthesis of 3,4,5,6-tetrasubstituted 3,4-dihydro-2H-pyrans (**346**) via a K_2CO_3 -mediated Michael-type cyclization reaction, as delineated in Scheme 204. Recently, they have synthesized 3-benzyl-2-hydroxy-7,8-dihydro-6H-quinolin-5-ones (**347**) via a reaction between the Baylis—Hillman acetates and cyclic enaminones in moderate yields (Scheme 205). Subsequently, they have also described the unusual formation of fully substituted furan derivatives (**348**) from 3-aryl-2-methylene-4-nitroalkanoates under the influence of TFA and H_2SO_4 (Scheme 206).

Kim et al. described a regioselective synthesis of tetrasubstituted pyrazole derivatives (**349**) from the reaction of the Baylis—Hillman adducts of methyl vinyl ketone and hydrazine hydrochloride under conventional heating (Scheme 207).²⁴³ Subsequently, Mamaghani et al. prepared the same pyrazoles under microwave irradiation.²⁴⁴

The transformation of cinnamyl bromides to cyclopropane derivatives (350) in moderate yields was reported by Kim et al. via a sequential in situ generation of the sulfur ylide

and reaction with methyl vinyl ketone or 2-chloroacrylonitrile (Scheme 208). 245

Liu et al. described an efficient BF₃·OEt₂-catalyzed C-C coupling reaction between α -EWG ketene-(S,S)-acetals and various Baylis-Hillman alcohols, which led to the synthesis of unsymmetrical biaryls (**351**) from the 1,4-pentadienes by a one-pot annulation-aromatization process (Scheme 209).

Very recently, Morizur and Mathias described the synthesis of polyfunctional 2-pyrrolidinones such as (352) from methyl 2-(carboethoxyhydroxymethyl)acrylate via a Michael addition—cyclization reaction, as delineated in Scheme 210.²⁴⁷

Lee et al. reported the preparation of methyl 2-amino-3*H*-1-benzazepine-4-carboxylates (**354**) or methyl 2-(cyanomethyl)-2,3-dihydro-1*H*-indole-2-carboxylates (**355**) from the reaction of 3-[2-formamido-, 2-acetamido- or 2-(propanoylamino)-phenyl]-substituted methyl 2-(cyanomethyl)propenoates (**353**) with sodium methoxide in methanol. The latter substrate was readily synthesized from the Morita—Baylis—Hillman reaction

Scheme 204.

Scheme 205.

OH COMe HBr R
$$\frac{1}{|U|}$$
 R = H, 4-Me, 4-Cl $\frac{OH}{Ar}$ R $\frac{OH}{U}$ R $\frac{OH}{Br}$ R $\frac{DABCO}{NO_2CH_2CO_2Et}$ R $\frac{EtO_2C}{O}$ $\frac{NO_2}{O}$ $\frac{ArH, 60-70 \text{ °C}, 2 \text{ h}}{35-56\%}$ R $\frac{EtO_2C}{ArH, 60-70 \text{ °C}, 2 \text{ h}}$ $\frac{Ar}{Ar}$ Ar = Ph, 2,5-Me₂-C₆H₃, 2,4,6-Me₃-C₆H₂ $\frac{348}{Ar}$

Scheme 206.

Scheme 207. Scheme 208.

Scheme 209.

Scheme 210.

of *N*-protected 2-aminobenzaldehydes with methyl acrylate followed by acetylation and cyanation (Scheme 211). ²⁴⁸

Song and Lee reported a facile strategy for the synthesis of pyrrolo[2,1-*b*]thiazoles (357) involving the thermal cyclization of several acetyl derivatives of the Baylis—Hillman adducts of thiazole-2-carbaldehydes (356). The reaction was also successful with the Morita—Baylis—Hillman acetates (358) obtained from 2-cyclopentenone and 2-cyclohexenone to yield the corresponding 5,6-dihydro-7*H*-cyclopenta[*b*]pyrrolo[2,1-*b*]-thiazol-7-one and 6,7-dihydrothiazolo[3,2-*a*]indol-8(5*H*)-one (359), as shown in Scheme 212.²⁴⁹

Marko et al. demonstrated that bicyclic enediones (**360**) of various sizes can be efficiently assembled by intramolecular Stetter cyclization of the readily available acetyl derivatives of Morita—Baylis—Hillman adducts of cyclic enones and 4-pentenal, as shown in Scheme 213.²⁵⁰

Tang et al. described an exciting Ph₃P-catalyzed intramolecular ylide annulation of an allyl bromide derivative (**361**) for the construction of benzobicyclo[4.3.0] ring systems (**362** and **363**) with three continuous stereogenic centers in a single step. It was demonstrated that the same starting substrates could yield selectively benzobicyclo[4.3.0] products **362** or

Scheme 212.

Scheme 213.

363 by simply changing the base. By the use of Na₂CO₃, compound **362** was obtained as the major product, while Cs₂CO₃ yielded the product **363**. These workers also demonstrated that **362** isomerizes to **363** under the influence of a strong base and studied this transformation via a detailed ¹H NMR experiment (Scheme 214). Mechanistically, it was proposed that Ph₃P reacts with the bromide **361** to form an intermediate phosphonium salt, which was deprotonated by Cs₂CO₃, leading to the corresponding phosphonium ylide in situ. An intramolecular Michael addition of the ylide, followed by a Michael addition of the phosphonium salt and subsequent β-elimination of Ph₃P, afforded the required product.²⁵¹

Jorgensen et al. reported an unusual diarylprolinol ether-catalyzed tandem reaction of α,β -unsaturated aldehydes and Nazarov reagent following a Michael/Morita—Baylis—

Hillman mechanism for the synthesis of various types of optically active cyclohexenone (**364**) and cyclohexanone (**365**) derivatives with up to four stereocenters (Scheme 215). 252

Kim et al. utilized acyloxiranes (366) generated from the Baylis—Hillman adducts for the facile synthesis of 1,3,4-trisubstituted pyrazoles (367) via a reaction with hydrazine derivatives in dichloromethane, as shown in Scheme 216.²⁵³

Schiesser et al. also utilized the epoxide afforded by the Baylis—Hillman adducts to develop the synthesis of selenophene-3-carboxylic acid (368), as delineated in Scheme 217. This selenophene-3-carboxylic acid was further employed as substrate for obtaining selenomilfasartan, which proved to be a potent AT₁-receptor antagonist.²⁵⁴

Kim et al. transformed the Baylis-Hillman acetates into 2,4,5-trisubstituted pyrimidines (369) by reacting them with

367

Scheme 214.

$$R_{1} = \text{Et, t-Bu, allyl} \\ R_{2} = \text{Ph, 4-Cl-C}_{6}H_{4}, 4-\text{OMe-C}_{6}H_{4}, 4-\text{NO}_{2}\text{-C}_{6}H_{4}, \\ \text{CO}_{2}\text{Et, Et, (Z)-hex-3-enyl, 2-thienyl, 2-furyl} \\ \text{TolO}_{2}\text{S} \\ \text{O} \\ \text{O} \\ \text{CO}_{2}\text{Et} \\ \text{TolSO}_{2}\text{Na,} \\ \text{AcOH, EtOH, rt} \\ \text{51-71}\% \\ \text{OH} \\ \text{OH} \\ \text{O} \\ \text$$

Scheme 216.

EWG

`.O.

366

EWG = CO_2Me/Et , CN; R = Ph, 2,4- Cl_2 - C_6H_3 , 2,4- $(NO_2)_2$ - C_6H_3

MeCN

Ar = Ph, 4-Cl-C₆H₄, 4-NO₂-C₆H₄

CI(CH₂)₂CI, reflux, 26-72 h

36-75%

Scheme 217.

OAC
$$R_1$$
 EWG + R_2 NH reflux, 3-6 h R_1 = Ph, 4-Me-C₆H₄, 1-naphthyl, n-C₅H₁₁; R_2 = Ph, Me R_1 R_2 R_3 EWG + R_4 R_2 NH reflux, 3-6 h R_4 R_5 R_5 R_6 R_7 R_8 R_9 R

Scheme 218.

amidines, as delineated in Scheme 218. They reported that the yields of the substituted pyrimidines were better in the case of the derivatives of acrylates, as compared to those afforded by methyl vinyl ketone or acrylonitrile.²⁵⁵

Lee et al. reported synthesis of 3-carbomethoxy-2H-thiochromenes (370) from the acetyl derivatives of the Baylis—Hillman adducts by reacting them with Na₂S in DMSO/H₂O, as shown in Scheme 219. 256

Scheme 219.

Very recently, Sa et al. reported a facile synthesis of 1,3-thiazin-4-ones (371) in high yields from the allyl bromides via a reaction of thiourea in the presence of a base under an aqueous medium (Scheme 220).²⁵⁷

Scheme 220.

21. Natural products

The functional attributes of the Baylis—Hillman adducts and their derivatives make them appropriate precursors to several complex natural products. This has also been elegantly exemplified by several research workers prior to 2003. ²⁵⁸

Trost et al. have reported the DYKAT of Baylis—Hillman derivatives for the total synthesis of furaquinocin E. 259 They elegantly extended the scope of their strategy by developing the synthesis of furaquinocins A and B and three more analogs of furaquinocin E. 260 Their work highlighted the ability to use racemic Baylis—Hillman adducts for asymmetric synthesis (Schemes 221–223). Later, they successfully employed aliphatic alcohols as competent nucleophiles in the Pd-catalyzed DYKAT reactions, the utility of which was demonstrated via a concise total synthesis of the gastrulation inhibitor, (+)-hippospongic acid A, as shown in Scheme 224. 261

Krishna et al. successfully accomplished the total synthesis of syributins 1 and 2 using the Baylis—Hillman adduct of 2,3-*O*-isopropylidene-*R*-glyceraldehyde (372) as the starting material followed by ring-closing metathesis of the acrylate derivative of the resulting diol as the key step (Scheme 225).²⁶²

Almeida and Coelho demonstrated a stereoselective synthesis of N-Boc dolaproine (Dap, 373), an amino acid residue of the antineoplastic pentapeptide, dolastatin 10.263 Their synthesis included a Baylis-Hillman reaction between N-Boc-prolinal (374) and methyl acrylate, followed by a diastereoselective double-bond hydrogenation and hydrolysis of the ester moiety. In order to establish the configuration they subjected the intermediate to cyclization via a sequential TFA and K₂CO₃mediated reaction to yield a lactam derivative (375) (Scheme 226). Later, Coelho et al. elegantly utilized the Baylis-Hillman adduct of 2-fluorobenzaldehyde and methyl acrylate, as a starting material for the straightforward, enantioselective synthesis of R-(+)-2-ethyl-2,3-dihydrofuran-2-carboxylic acid (376), the direct precursor of R-(+)-efaroxan, which is used for the treatment of neurodegenerative diseases (Alzheimer's or Parkinson's disease), migraine and type II (non-insulin dependent) diabetes (Scheme 227).²⁶⁴

Corey et al. utilized an intramolecular Baylis-Hillman reaction between keto and acrylamide for the cyclization to

Scheme 221.

Scheme 222.

Scheme 223.

TBDPSO

OAc

TBDPSO

MeO₂C

HO

1. [Pd(
$$\pi$$
-allyl)Cl]₂, (R,R)-Ln

N(Hex)₄Cl, dioxane

2. LiOH, THF/H₂O

Ph
Ph
Ph
ONH
HN

(+)-hippospongic acid

Scheme 224.

DABCO, dioxane/
OEt
$$\frac{\text{DABCO, dioxane/}}{72\%}$$
OH $\frac{\text{CH}_2\text{Cl}_2, 0 °C to rt, 10 h}{75\%}$

OH $\frac{\text{CH}_2\text{Cl}_2, 0 °C to rt, 10 h}{75\%}$

OH $\frac{\text{CH}_2\text{Cl}_2, 0 °C to rt, 10 h}{75\%}$

OH $\frac{\text{CH}_2\text{Cl}_2, rt, 0.5 h}{87-90\%}$
OH $\frac{\text{CH}_2\text{Cl}_2, rt, 0.5 h}{87-90\%}$
OH $\frac{\text{CH}_2\text{Cl}_2, rt, 0.5 h}{86-90\%}$

Scheme 225.

Scheme 226.

Scheme 227.

produce a highly substituted 4-methylene-5-oxo-pyrrolidine derivative (**377**), which served as the starting material in the enantioselective total synthesis of salinosporamide A (Scheme 228).²⁶⁵ As this key step was time consuming, they developed an attractive method using a Kulinkovich reagent followed by treatment with I₂ and, later, with Et₃N to yield the diastereomerically pure pyrrolidinone (**378**).²⁶⁶ This was further utilized for the synthesis of a salinosporamide analog (Scheme 229).

Chang et al. reported the synthesis of N-alkyl 3-(E)-alkylidene-5-substituted sulfonylpiperidine-2,6-diones (379) via a [3+3] annulation strategy, which was used for a one-pot formal synthesis of tacamonine alkaloid, a vasodilator and hypotensive agent (Scheme 230). 267

Tadano et al. reported a total synthesis of natural (+)-tube-lactomicin A, a 16-membered macrolide antibiotic, which comprised 54 total steps from methyl (*R*)-lactate (380) in 6.2% overall yield involving the Baylis—Hillman reaction as one of the key steps (Scheme 231). ²⁶⁸

Subsequently, Orena et al. described the synthesis of conformationally restricted analogs of (S)- β -homoserine (**381**) and (S)-aspartic acid (**382**) starting from chiral 3,4-*trans*-disubstituted pyrrolidin-2-ones (**383**), which, in turn, were obtained from the Baylis–Hillman adducts, as shown in Scheme 232. 269

Franck et al. described a convergent, highly stereoselective total synthesis of (—)-spinosyn A, a polyketide natural product

Scheme 229.

$$R_1 = S \\ 0 \\ NH \\ MeO \\ 0 \\ NH \\ MeO \\ 0 \\ R_3 \\ NaH, THF, 67 °C \\ 41-75\% \\ R_2 \\ R = OH, OAc; R_1 = 4-Me-C_6H_4; R_2 = Me, allyl, n-Pr, Bn, BnCH_2, Trypt; R_3 = Me, Ph$$

Scheme 230.

Scheme 231.

$$\begin{array}{c} \text{OH} \\ \text{EtO}_2\text{C} \\ \end{array} \\ \text{EtO}_2\text{C} \\ \end{array} \\ \begin{array}{c} \text{NH}_2 \\ \text{NH}_2 \\ \text{MeOH, rt to 60 °C} \\ \text{DBU, PhMe, rt} \\ \end{array} \\ \begin{array}{c} \text{R}_1 \\ \text{N} \\ \text{CO}_2\text{Me} \\ \end{array} \\ \begin{array}{c} \text{Analogue of (S)-}\beta\text{-homoserine} \\ \text{BocHN} \\ \text{N} \\ \text{383} \\ \end{array} \\ \begin{array}{c} \text{BocHN} \\ \text{N} \\ \text{381} \\ \text{Solution of (S)-}\beta\text{-homoserine} \\ \end{array} \\ \begin{array}{c} \text{R}_1 \\ \text{N} \\$$

Scheme 232.

possessing an extraordinary insecticidal activity. The key feature of the synthesis included a transannular Diels—Alder reaction of a macrocyclic pentaene (**384**) and transannular vinylogous Morita—Baylis—Hillman cyclization to yield (—)-spinosin A (Scheme 233). 270 In addition, they successfully synthesized $C_1 - C_{11}$ the fragment (**385**) of caribenolide I, a 26-membered macrolactone that possesses in vitro cytotoxicity against human colon tumor cells, starting from the Baylis—Hillman reaction between 3-para-methoxybenzyloxypropanal and methyl acrylate in the presence of 3-hydroxyquinuclidine, as illustrated in Scheme 234. 271

Rodgen and Schaus reported an efficient route for the construction of the clerodane decalin core (**386**) of asmarines A and B using an asymmetric Baylis—Hillman reaction followed by a Lewis acid-mediated ring-annulation strategy, as shown in Scheme 235.²⁷²

Recently, Jogireddy and Maier have developed a novel route for the synthesis of luminacin D. The starting aldehyde (387), which was obtained by a simple Baylis—Hillman reaction between acetaldehyde and methyl acrylate, followed by OH transposition, was extended by two highly stereoselective asymmetric aldol reactions (Scheme 236).²⁷³

Scheme 233.

Scheme 234.

Scheme 235.

Scheme 236.

Hiemstra et al. described the synthesis of the compact tricyclic core (**388**) of solanoeclepin A, a hatching agent of potato cyst nematodes. The total synthesis required the allenic bromide (**389**), which was prepared from the Baylis—Hillman adduct of allene (Scheme 237).²⁷⁴

Brase et al. have developed a synthetic route to tetrahy-droxanthenone mycotoxins (390) via a domino oxa-Michael—aldol condensation. By applying this methodology, they achieved the first total synthesis of the secondary metabolite,

diversonol, in a racemic form, as delineated in Scheme 238^{275}

Selvakumar et al. described an RCM reaction of electrondeficient dienes (391) for the synthesis of diverse butenolides (392) and employed this methodology in the total synthesis of (\pm)-phaseolinic acid, a natural product of the paraconic acid class (Scheme 239).²⁷⁶

Aggarwal et al. reported a novel methodology in which a broad range of Michael acceptors were allowed to couple

Scheme 237.

Scheme 238.

Scheme 239.

with the readily available iminium ion in an inter- and intra-molecular Morita—Baylis—Hillman-type reaction to afford densely functionalized heterocycles. The iminium ions generally present as masked N,O-acetals were generated by TMSOTf, while BF₃·Et₂O in the presence of Me₂S was used to accomplish the reaction. More importantly, the process was highly enantioselective for cyclic enones (**393**). By employing this methodology, they reported a short synthesis of (+)-heliotridine, as shown in Scheme 240.

Recently, Murphy et al. reported the formation of a side product **397** resulting from a Baylis—Hillman-type cyclization during the preparation of the diene **396** from the *tert*-butyl acetate **394** and the aldehyde **395** in their endeavors to carry out a Knoevenagel condensation between **394** and **395** en route to the total synthesis of crambescidin **359** (Scheme **241**). ²⁷⁸

Koert et al. utilized the Baylis—Hillman adduct to generate the molecular framework for the efficient stereoselective synthesis of methyl 7-dihydro-trioxacarcinoside B (**398**). The key steps in the process were biocatalytic resolution of the Baylis—Hillman adduct, RCM reaction, a substrate-controlled epoxidation, and stereo- and regio-controlled opening of epoxide by allyl alcohol, as shown in Scheme 242. ²⁷⁹

Ciufolini et al. utilized the Baylis—Hillman adducts to synthesize ketones **400**, via trialkylsilyl triflate/Et₃N-catalyzed cyclization, which served as useful building blocks for the preparation of analogs of the potent antifungal agent, sordarin. It was presumed that the exposure to TIPSOTf induces the

formation of the bis-trialkylsilyl derivative **399**, which undergoes a spontaneous Diels—Alder reaction to furnish the expected product as mixture of diastereomers (Scheme 243).²⁸⁰

Other natural products having open chain structures have also been synthesized employing Baylis—Hillman chemistry. These include (2*E*)-2-butyloct-2-enal, ²⁸¹ *cis*-hedione and methyl jasmonate, ²⁸² 4,5-dihydroxy-2,3-pentanedione (DPD), ²⁸³ and 5-*O*-acylated derivatives of DPD, ²⁸⁴ sitophilate, ²⁸⁵ (+)-(*S*)-manicone, ²⁸⁶ (+)-(*S*)-normanicone, ²⁸⁷ (+)-(*S*)-dominicalure-I and (+)-(*S*)-dominicalure-II, ²⁸⁸ 1-[*p*-(myristyloxy)- α -methylcinnamoyl]glycerol (LK-903), ²⁸⁹ F-3-2-5, ²⁹⁰ and umbelactones. ²⁹¹ These natural products are, however, not discussed here in detail, as ther are beyond the scope of this review.

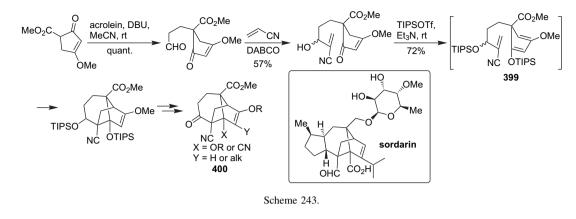
22. Conclusions

The expounding of the synthetic applications of the Baylis—Hillman adducts and their derivatives for the generation of cyclic compounds, besides a variety of other products, clearly establishes that this reaction has become a standard synthetic methodology in the arsenal of organic chemists. Figure 10 summarizes the different reaction strategies, which have been discussed in this article for generating the cyclic compounds. Although tremendous advances have been achieved in this field, we firmly believe that the applications of the Baylis—Hillman derivatives to accomplish the synthesis of novel prototypes will continue to grow. Such growth will

Scheme 240.

Scheme 241.

methyl 7-dihydro-trioxacarcinoside B



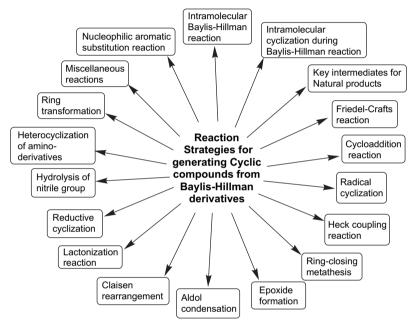


Figure 10. Reaction strategies employed on the Baylis-Hillman derivatives for the generation of the cyclic compounds.

surely rely on discovering intelligent applications of the already reported strategies with little variations. Nevertheless, with a view to enhance the development with respect to cyclic frameworks, there are still many opportunities, which have not been explored explicitly. Some of these have been highlighted in this report as they might motivate researchers across the boundaries.

The Baylis—Hillman derivatives originating from substituted heterocyclic aldehydes are excellent synthons for the synthesis of hetero-annulated or polycyclic ring systems bearing a resemblance either to natural products or bioactive molecules. Except for a few isolated reports, however, not many efforts have been made to generate cyclic frameworks in this fashion. It is likely that the behavior of the formyl group within the heterocycle for the Baylis—Hillman reaction might be an issue. Intriguingly, the formyl groups present at different positions in a heterocyclic aldehyde show distinctly different rates for this reaction and, to date, no experimental evidence has been provided to establish a reason behind such behavior.

There are still only a limited number of fast-reacting electrophiles discovered for the Baylis-Hillman reaction. On

analysis of the publications related to the development of a new catalytic system or medium, it has been noted that most of the reports exemplify the case of substituted nitrobenz-aldehydes, which are already known to be fast-reacting substrates for the Baylis—Hillman reaction. Hence, any such exercise would be more fruitful if the investigations are carried out over a range of substrates. The solid-phase methodologies for the Baylis—Hillman reaction are mostly limited to reactions of the acrylate resins with aldehydes. Due to the slow rate of reaction of different electrophiles, the building up of chemical libraries employing this reaction is cumbersome and the products are generally associated with impurities.

The synthetic potential of some of the intermediates, which could be efficiently generated through this chemistry needs to be explored further. As an example, the *N*-formamide derivative generated from the primary allyl amines could be employed for the generation of isocyanides, which may, in turn, undergo different multicomponent reactions. Likewise, the Baylis—Hillman derivatives of sugar-based compounds could be effectively utilized for the generation of several

sugar-heterocycle hybrid molecules for development of novel enzyme inhibitors.

To date, there have been more reports of the synthetic applications of the Baylis—Hillman chemistry rather than the utility of the generated products in various spheres of life. Probably due to this reason, it has not been considered essential to investigate the reported synthetic strategies for cyclization at higher scales. Therefore, it would be fruitful to discover the bio-properties in the plethora of compounds, which have already been generated in this fashion and also examine the applications of these reaction strategies at higher scales.

Nevertheless, it can be extrapolated from the tremendous advances in the applications of the Baylis—Hillman derivatives that the fruits of such an exercise will ultimately lead to privileged compounds, which would be fine tuned to the advantage of mankind.

Acknowledgements

One of the authors (V.S.) gratefully acknowledges the financial support from the CSIR, New Delhi in the form of a fellowship. We acknowledge the fruitful discussions carried out with Dr. A. P. Bhaduri during compilation of this manuscript. The work on synthesis of cyclic compounds via Baylis—Hillman chemistry in our laboratory is funded by DST, New Delhi.

References and notes

- (a) Schreiber, S. L. Science 2000, 287, 1964; (b) Burke, M. D.; Berger, E. M.; Schreiber, S. T. Science 2003, 302, 613.
- 2. Basavaiah, D.; Rao, K. V.; Reddy, R. J. Chem. Soc. Rev. 2007, 36, 1581.
- (a) Morita, K. Japan Patent 6803364, 1968; Chem. Abstr. 1968, 69, 58828s;
 (b) Morita, K.; Suzuki, Z.; Hirose, H. Bull. Chem. Soc. Jpn. 1968, 41, 2815.
- Baylis, A. B.; Hillman, M. E. D.; German Patent 2155113, 1972; Chem. Abstr. 1972, 77, 34174q.
- 5. Drewes, S. E.; Emslie, N. D. J. Chem. Soc., Perkin Trans. 1 1982, 2079.
- Hoffmann, H. M. R.; Rabe, J. Angew. Chem., Int. Ed. Engl. 1983, 22, 795
- 7. Drewes, S. E.; Roos, G. H. P. Tetrahedron 1988, 44, 4653.
- 8. Basavaiah, D.; Rao, P. D.; Hyma, R. S. Tetrahedron 1996, 52, 8001.
- 9. Ciganek, E. Organic Reactions; Paquette, L. A., Ed.; John Wiley & Sons: New York, NY, 1997; Vol. 51, p 201.
- 10. Basavaiah, D.; Rao, A. J.; Satyanarayana, T. Chem. Rev. 2003, 103, 811.
- 11. Langer, P. Angew. Chem., Int. Ed. 2000, 39, 3049.
- 12. Kim, J. N.; Lee, K. Y. Curr. Org. Chem. 2002, 6, 627.
- (a) Shi, Y.-L.; Shi, M. Eur. J. Org. Chem. 2007, 2905; (b) Shi, Y.-L.; Shi, M. Org. Biomol. Chem. 2007, 5, 1499.
- Masson, G.; Housseman, C.; Zhu, J. Angew. Chem., Int. Ed. 2007, 46, 4614.
- 15. See Ref. 6.
- (a) Hill, J. S.; Isaacs, N. S. Tetrahedron Lett. 1986, 27, 5007; (b) Hill,
 J. S.; Isaacs, N. S. J. Phys. Org. Chem. 1990, 3, 285.
- 17. Bode, M. L.; Kaye, P. T. Tetrahedron Lett. 1991, 32, 5611.
- Price, K. E.; Broadwater, S. J.; Jung, H. M.; McQuade, D. T. Org. Lett. 2005, 7, 147.
- Price, K. E.; Broadwater, S. J.; Walker, B. J.; McQuade, D. T. J. Org. Chem. 2005, 70, 3980.
- 20. Xu, J. Theochem 2006, 767, 61.
- 21. Roy, D.; Sunoj, R. B. Org. Lett. 2007, 9, 4873.
- Robiette, R.; Aggarwal, V. K.; Harvey, J. N. J. Am. Chem. Soc. 2007, 129, 15513.

- (a) Black, G. P.; Dinon, F.; Fratucello, S.; Murphy, P. J.; Nielsen, M.; Williams, H. L.; Walshe, N. D. A. *Tetrahedron Lett.* 1997, 38, 8561;
 (b) Dinon, F.; Richards, E.; Murphy, P. J.; Hibbs, D. E.; Hursthouse, M. B.; Malic, K. M. A. *Tetrahedron Lett.* 1999, 40, 3279;
 (c) Richards, E. L.; Murphy, P. J.; Dinon, F.; Fratucello, S.; Brown, P. M.; Gelbrich, T.; Hursthouse, M. B. *Tetrahedron* 2001, 57, 7771.
- 24. Methot, J. L.; Roush, W. R. Org. Lett. 2003, 5, 4223.
- 25. Yeo, J. E.; Yang, X.; Kim, H. J.; Koo, S. Chem. Commun. 2004, 236.
- Krishna, P. R.; Kannan, V.; Sharma, G. V. M. J. Org. Chem. 2004, 69, 6467.
- 27. Aroyan, C. E.; Vasbinder, M. M.; Miller, S. J. Org. Lett. 2005, 7, 3849.
- 28. Seidel, F.; Gladysz, J. A. Synlett 2007, 986.
- Jellerichs, B. G.; Kong, J.-R.; Krische, M. J. J. Am. Chem. Soc. 2003, 125, 7758.
- 30. Krafft, M. E.; Haxell, T. F. N. J. Am. Chem. Soc. 2005, 127, 10168.
- Krafft, M. E.; Haxell, T. F. N.; Seibert, K. A.; Hirosawa, C. Chem. Commun. 2005, 5772.
- Krafft, M. E.; Haxell, T. F. N.; Seibert, K. A.; Abboud, K. A. J. Am. Chem. Soc. 2006, 128, 4174.
- 33. Krafft, M. E.; Seibert, K. A. Synlett 2006, 3334.
- 34. Krafft, M. E.; Wright, J. A. Chem. Commun. 2006, 2977.
- 35. Chen, S.-H.; Hong, B.-C.; Sua, C.-F.; Sarshar, S. Tetrahedron Lett. 2005, 46, 8899.
- Pigge, F. C.; Dhanya, R.; Hoefgen, E. R. Angew. Chem., Int. Ed. 2007, 46, 2887.
- 37. Basavaiah, D.; Rao, A. J. Chem. Commun. 2003, 604.
- 38. Basavaiah, D.; Rao, A. J. Tetrahedron Lett. 2003, 44, 4365.
- 39. Coelho, F.; Diag, G.; Abella, C. A. M.; Almeida, W. P. Synlett 2006, 435.
- (a) Clive, D. L. J.; Yu, M.; Li, Z. Chem. Commun. 2005, 906; (b) Clive,
 D. L. J.; Li, Z.; Yu, M. J. Org. Chem. 2007, 72, 5608.
- 41. Virieux, D.; Guillouzic, A.-F.; Cristau, H.-J. Tetrahedron 2006, 62, 3710.
- 42. Song, Y. S.; Lee, C. H.; Lee, K.-J. J. Heterocycl. Chem. 2003, 40, 939.
- 43. Yi, H.-W.; Park, H. W.; Song, Y. S.; Lee, K.-J. Synthesis 2006, 1953.
- 44. Pan, W.; Dong, D.; Sun, S.; Liu, Q. Synlett 2006, 1090.
- 45. Lee, K. Y.; Kim, J. M.; Kim, J. N. Synlett 2003, 357.
- 46. Kaye, P. T.; Musa, M. Y.; Nocanda, X. W. Synthesis 2003, 531.
- 47. Kaye, P. T.; Musa, M. A. Synth. Commun. 2003, 33, 1755.
- 48. Lee, K. Y.; Kim, J. M.; Kim, J. N. Bull. Korean Chem. Soc. 2003, 24, 17.
- 49. Lesch, B.; Brase, S. Angew. Chem., Int. Ed. 2004, 43, 115.
- Nising, C. F.; Ohhemuller, U. K.; Friedrich, A.; Lesch, B.; Steiner, J.;
 Schnockel, H.; Nieger, M.; Brase, S. Chem.—Eur. J. 2006, 12, 3647.
- Ohhemuller, U. K.; Nising, C. F.; Nieger, M.; Brase, S. Eur. J. Org. Chem. 2006, 1535.
- 52. Shi, Y.-L.; Shi, M. Synlett 2005, 2623.
- 53. Shi, Y.-L.; Shi, M. Org. Lett. 2005, 7, 3057.
- 54. Guo, Y.-W.; Shi, Y.-L.; Li, H.-B.; Shi, M. Tetrahedron 2006, 62, 5875.
- 55. Shi, Y.-L.; Shi, M. Chem.—Eur. J. 2006, 12, 3374.
- 56. Qi, M.-J.; Shi, M. Tetrahedron 2007, 63, 10415.
- Shah, B. A.; Taneja, S. C.; Sethi, V. K.; Gupta, P.; Andotra, S. S.; Chimnib, S. S.; Qazi, G. N. *Tetrahedron Lett.* 2007, 48, 955.
- Rios, R.; Sunden, H.; Ibrahem, I.; Zhao, G.-L.; Cordova, A. *Tetrahedron Lett.* 2006, 47, 8679.
- Rios, R.; Sunden, H.; Ibrahem, I.; Zhao, G.-L.; Eriksson, L.; Cordova, A. Tetrahedron Lett. 2006, 47, 8547.
- (a) Wang, W.; Li, H.; Wang, J.; Zu, L. J. Am. Chem. Soc. 2006, 128, 10354;
 (b) Zu, L.; Wang, J.; Li, H.; Xie, H.; Jiang, W.; Wang, W. J. Am. Chem. Soc. 2007, 129, 1036.
- Kinoshita, H.; Kinoshita, S.; Munechika, Y.; Iwamura, T.; Watanabe, S.-i.; Kataoka, T. Eur. J. Org. Chem. 2003, 24, 4852.
- (a) Kim, J. N.; Lee, H. J.; Lee, K. Y.; Kim, H. S. Tetrahedron Lett. 2001,
 42, 3737; (b) Chung, Y. M.; Lee, H. J.; Hwang, S. S.; Kim, J. N. Bull.
 Korean Chem. Soc. 2001, 22, 799.
- 63. (a) Kim, J. N.; Kim, H. S.; Gong, J. H.; Chung, Y. M. Tetrahedron Lett. 2001, 42, 8341; (b) Kim, J. N.; Im, Y. J.; Gong, J. H.; Lee, K. Y. Tetrahedron Lett. 2001, 42, 4195; (c) Im, Y. J.; Chung, Y. M.; Gong, J. H.; Kim, J. N. Bull. Korean Chem. Soc. 2002, 23, 787.
- 64. Hong, W. P.; Lee, K.-J. Synthesis 2005, 33.
- 65. Horn, C. R.; Perez, M. Synlett 2005, 1480.

- 66. Hong, W. P.; Lee, K.-J. Synthesis 2006, 963.
- Narender, P.; Srinivas, U.; Ravinder, M.; Rao, B. A.; Ramesh, C.; Harakishore, K.; Gangadasu, B. U.; Murthy, S. N.; Rao, V. J. Bioorg. Med. Chem. 2006, 14, 4600.
- (a) Basavaiah, D.; Pandiaraju, S.; Padmaja, K. Synlett 1996, 393;
 (b) Basavaiah, D.; Krishnamacharyulu, M.; Suguna Hyma, R.; Pandiaraju, S. Tetrahedron Lett. 1997, 38, 2141.
- (a) Basavaiah, D.; Bakthadoss, M.; Jayapal Reddy, G. Synthesis 2001, 919;
 (b) Basavaiah, D.; Mallikarjuna Reddy, R. Tetrahedron Lett. 2001, 42, 3025;
 (c) Lee, H. J.; Kim, T. H.; Kim, J. N. Bull. Korean Chem. Soc. 2001, 22, 1063;
 (d) Basavaiah, D.; Bakthadoss, M.; Pandiaraju, S. Chem. Commun. 1998, 1639;
 (e) Roy, O.; Riahi, A.; Henin, F.; Muzart, J. Tetrahedron 2000, 56, 8133.
- 70. Basavaiah, D.; Satyanarayana, T. Chem. Commun. 2004, 32.
- Basavaiah, D.; Sharada, D. S.; Veerendhar, A. Tetrahedron Lett. 2004, 45, 3081.
- Das, B.; Majhi, A.; Banerjee, J.; Chowdhury, N.; Holla, H.; Harakishore, K.; Murty, U. S. Chem. Pharm. Bull. 2006, 54, 403.
- GowriSankar, S.; Lee, K. Y.; Lee, C. G.; Kim, J. N. Tetrahedron Lett. 2004, 45, 6141.
- Lee, C. G.; Lee, K. Y.; GowriSankar, S.; Kim, J. N. Tetrahedron Lett. 2004, 45, 7409.
- Gowrisankar, S.; Lee, K. Y.; Kim, J. N. Bull. Korean Chem. Soc. 2005, 26, 1112.
- 76. Shanmugam, P.; Rajasingh, P. Chem. Lett. 2005, 34, 1494.
- Hong, W. P.; Lim, H. N.; Park, H. W.; Lee, K.-J. Bull. Korean Chem. Soc. 2005, 26, 655.
- 78. Lee, K. Y.; Kim, S. C.; Kim, J. N. Tetrahedron Lett. 2006, 47, 977.
- 79. Lee, K. Y.; Seo, J.; Kim, J. N. Tetrahedron Lett. 2006, 47, 3913.
- 80. Lim, H. N.; Ji, S.-H.; Lee, K. J. Synthesis 2007, 2454.
- Das, B.; Chowdhury, N.; Damodar, K.; Benerjee, J. Chem. Pharm. Bull. 2007, 55, 1274.
- Rajan, Y. C.; Kanakam, C. C.; Selvam, S. P.; Murugesan, K. Tetrahedron Lett. 2007, 48, 8562.
- (a) Kanemasa, S.; Kobayashi, S. Bull. Chem. Soc. Jpn. 1993, 66, 2685;
 (b) Micuch, P.; Fisera, L.; Cyranski, M. K.; Krygowski, T. M. Tetrahedron Lett. 1999, 40, 167;
 (c) Micuch, P.; Fisera, L.; Cyranski, M. K.; Krygowski, T. M.; Krajcik, J. Tetrahedron 2000, 56, 5465.
- Dugovic, B.; Fisera, L.; Hametner, C.; Cyranski, M. K.; Pronayova, N. Monatsh. Chem. 2004, 135, 685.
- 85. Dugovic, B.; Fisera, L.; Hametner, C.; Pronayova, N. ARKIVOC 2003,
- 86. Das, B.; Mahender, G.; Holla, H.; Banerjee, J. ARKIVOC 2005, 27.
- 87. Jayashankaran, J.; Manian, R. D. R. S.; Sivaguru, M.; Raghunathan, R. Tetrahedron Lett. 2006, 47, 5535.
- Ramesh, E.; Kathiresan, M.; Raghunathan, R. Tetrahedron Lett. 2007, 48, 1835.
- Sreedhar, B.; Reddy, P. S.; Kumar, N. S. Tetrahedron Lett. 2006, 47, 3055.
- Chandrasekhar, S.; Basu, D.; Rambabu, C. Tetrahedron Lett. 2006, 47, 3059.
- 91. Cho, H. I.; Lee, S. W.; Lee, K. J. J. Heterocycl. Chem. 2004, 41, 799.
- 92. Ko, S. H.; Lee, K. J. J. Heterocycl. Chem. 2004, 41, 613.
- 93. Song, Y. S.; Lee, K.-J. J. Heterocycl. Chem. 2006, 43, 1721.
- Ji, S. H.; Hong, W. P.; Ko, S.-H.; Lee, K. J. J. Heterocycl. Chem. 2006, 43, 799.
- Shanmugam, P.; Viswambharan, B.; Madhavan, S. Org. Lett. 2007, 9, 4095
- Bakthadoss, M.; Sivakumar, N.; Sivakumar, G.; Murugan, G. Tetrahedron Lett. 2007, 49, 820.
- 97. Ramesh, E.; Raghunathan, R. Tetrahedron Lett. 2008, 49, 1125.
- 98. Aggarwal, V. K.; Patin, A.; Tisserand, S. Org. Lett. 2005, 7, 2555.
- 99. Nair, V.; Abhilash, K. G. Synthesis 2005, 1967.
- 100. Du, Y.; Lu, X.; Zhang, C. Angew. Chem., Int. Ed. 2003, 42, 1035.
- 101. Du, Y.; Feng, J.; Lu, X. Org. Lett. 2005, 7, 1987.
- 102. Feng, J.; Lu, X.; Kong, A.; Han, X. Tetrahedron 2007, 63, 6035.
- 103. Mix, S.; Blechert, S. Org. Lett. 2005, 7, 2015.
- 104. Song, H. S.; Song, Y. S.; Lee, K.-J. J. Heterocycl. Chem. 2006, 43, 1533.

- Kobbelgaard, S.; Brandes, S.; Jorgensen, K. A. Chem.—Eur. J. 2008, 14, 1464
- Park, D. Y.; Kim, S. J.; Kim, T. H.; Kim, J. N. Tetrahedron Lett. 2006, 47, 6315.
- Kim, S. C.; Lee, H. S.; Lee, Y. J.; Kim, J. N. Tetrahedron Lett. 2006, 47, 5681
- 108. (a) Lee, M. J.; Lee, K. Y.; Gowrisankar, S.; Kim, J. N. Tetrahedron Lett. 2006, 47, 1355; (b) Lee, M. J.; Lee, K. Y.; Park, D. Y.; Kim, J. N. Tetrahedron 2006, 62, 3128.
- Park, D. Y.; Gowrisankar, S.; Kim, J. N. Tetrahedron Lett. 2006, 47, 6641
- 110. Park, D. Y.; Lee, K. Y.; Kim, J. N. Tetrahedron Lett. 2007, 48, 1633.
- 111. Zhao, G.-L.; Huang, J.-W.; Shi, M. Org. Lett. 2003, 5, 4737.
- 112. Shi, M.; Zhao, G.-L. Adv. Synth. Catal. 2004, 346, 1205.
- 113. Zhao, G.-L.; Shi, M. J. Org. Chem. 2005, 70, 9975.
- 114. Shi, M.; Zhao, G.-L. Org. Biomol. Chem. 2005, 3, 3686.
- 115. Kim, S. C.; Lee, K. Y.; Lee, H. S.; Kim, J. N. Tetrahedron 2008, 64, 103.
- 116. Shanmugam, P.; Rajasingh, P. Tetrahedron 2004, 60, 9283.
- 117. Shanmugam, P.; Rajasingh, P. Synlett 2005, 939.
- 118. Shanmugam, P.; Rajasingh, P. Tetrahedron Lett. 2005, 46, 3369.
- Shanmugam, P.; Rajasingh, P.; Viswambharan, B.; Vaithiyanathan, V. Synth. Commun. 2007, 37, 2291.
- (a) Gowrisankar, S.; Lee, K. Y.; Kim, J. N. Tetrahedron Lett. 2005, 46, 4859; (b) Gowrisankar, S.; Kim, S. J.; Kim, J. N. Tetrahedron Lett. 2007, 48, 289.
- Gowrisankar, S.; Lee, K. Y.; Kim, T. H.; Kim, J. N. Tetrahedron Lett. 2006, 47, 5785.
- Gowrisankar, S.; Lee, H. S.; Kim, J. N. Bull. Korean Chem. Soc. 2006, 27, 2097
- 123. Gowrisankar, S.; Lee, H. S.; Kim, J. N. Tetrahedron Lett. 2007, 48, 3105.
- 124. Park, D. Y.; Gowrisankar, S.; Kim, J. N. Bull. Korean Chem. Soc. 2005, 26, 1440.
- Gowrisankar, S.; Lee, K. Y.; Kim, J. N. Bull. Korean Chem. Soc. 2006, 27, 929.
- 126. Singh, V.; Batra, S. Tetrahedron Lett. 2006, 47, 7043.
- Majhi, T. P.; Neogi, A.; Ghosh, S.; Mukherjee, A. K.; Chattopadhyay, P. Tetrahedron 2006, 62, 12003.
- Gowrisankar, S.; Kim, S. J.; Lee, J.-E.; Kim, J. N. Tetrahedron Lett. 2007, 48, 4419.
- 129. Erkoc, S.; Mathias, L. J.; Acar, A. E. Macromolecules 2006, 39, 8936.
- Gowrisankar, S.; Lee, H. S.; Kim, J. N. Bull. Korean Chem. Soc. 2007, 28, 2501.
- Lee, H.-S.; Kim, H.-S.; Kim, J.-M.; Kim, J.-N. Tetrahedron 2008, 64, 2397.
- (a) Basavaiah, D.; Muthukumaran, K. Tetrahedron 1998, 54, 4943; (b)
 Sundar, N.; Bhat, S. V. Synth. Commun. 1998, 28, 2311; (c) Kumareswaran, R.; Vankar, Y. D. Synth. Commun. 1998, 28, 2291.
- Park, J. B.; Ko, S. H.; Hong, W. P.; Lee, K. J. Bull. Korean Chem. Soc. 2004, 25, 927.
- 134. Vasudevan, A.; Tseng, P. S.; Djuric, S. W. *Tetrahedron Lett.* **2006**, 47,
- Chen, B.; Xie, X.; Lu, J.; Wang, Q.; Zhang, J.; Tang, S.; She, X. Synlett 2006, 259.
- Gowrisankar, S.; Lee, H. S.; Lee, K. Y.; Lee, J.-E.; Kim, J. N. Tetrahedron Lett. 2007, 48, 8619.
- Gowrisankar, S.; Lee, H. S.; Kim, J. M.; Kim, J. N. Tetrahedron Lett. 2008, 49, 1670.
- 138. Paquette, L. A.; Mendez-Andino, J. Tetrahedron Lett. 1999, 40, 4301.
- Kim, J. M.; Lee, K. Y.; Lee, S.; Kim, J. N. Tetrahedron Lett. 2004, 45, 2805.
- 140. Lee, K. Y.; Na, J. E.; Lee, J. Y.; Kim, J. N. Bull. Korean Chem. Soc. 2004, 25, 1280.
- 141. Gowrisankar, S.; Lee, K. Y.; Kim, J. N. Tetrahedron 2006, 62, 4052.
- 142. Balan, D.; Adolfsson, H. Tetrahedron Lett. **2004**, 45, 3089.
- Declerck, V.; Ribiere, P.; Martinez, J.; Lamaty, F. J. Org. Chem. 2004, 69, 8372.
- Declerck, V.; Allouchi, H.; Martinez, J.; Lamaty, F. J. Org. Chem. 2007, 72, 1518.

- Krafft, M. E.; Song, E.-H.; Davoile, R. J. Tetrahedron Lett. 2005, 46, 6359
- 146. Chen, P.-Y.; Chen, H.-M.; Chen, L.-Y.; Tzeng, J.-Y.; Tsai, J.-C.; Chi, P.-C.; Lic, S.-R.; Wang, E.-C. *Tetrahedron* **2007**, *63*, 2824.
- 147. Krishna, P. R.; Narsingam, M. J. Comb. Chem. 2007, 9, 62.
- 148. Trost, B. M.; Brennan, M. K. Org. Lett. 2007, 9, 3961.
- Gowrisankar, S.; Kim, H. S.; Lee, H. S.; Kim, J. N. Bull. Korean Chem. Soc. 2007, 28, 1844.
- 150. Doddi, V. R.; Vankar, Y. D. Eur. J. Org. Chem. 2007, 5583.
- Donohoe, T. J.; Kershaw, N. M.; Orr, A. J.; Wheelhouse (nee Gosby),
 K. M. P.; Fishlock, L. P.; Lacy, A. R.; Bingham, M.; Procopiou, P. A.
 Tetrahedron 2008, 64, 809.
- Porto, R. S.; Vasconcellos, M. L. A. A.; Ventura, E.; Coelho, F. *Synthesis* 2005, 2297.
- 153. Raheem, I. T.; Jacobsen, E. N. Adv. Synth. Catal. 2005, 347, 1701.
- 154. Davoust, M.; Briere, J.-F.; Metzner, P. Org. Biomol. Chem. 2006, 4, 3048
- Das, B.; Holla, H.; Venkateswarlu, K.; Majhi, A. Tetrahedron Lett. 2005, 46, 8895.
- (a) Chamakh, A.; Amri, H. *Tetrahedron Lett.* **1998**, *39*, 375; (b)
 Chamakh, A.; M'hirsi, M.; Villieras, J.; Lebreton, J.; Amri, H. *Synthesis* **2000**, 295.
- Im, Y. J.; Lee, C. G.; Kim, H. R.; Kim, J. N. Tetrahedron Lett. 2003, 44, 2987
- 158. Kim, J. N.; Kim, J. M.; Lee, K. Y. Synlett 2003, 821.
- Kim, J. M.; Lee, K. Y.; Kim, J. N. Bull. Korean Chem. Soc. 2004, 25, 328
- Park, D. Y.; Lee, M. J.; Kim, T. H.; Kim, J. N. Tetrahedron Lett. 2005, 46, 8799.
- Liu, X.; Zhao, J.; Jin, G.; Zhao, G.; Zhu, S.; Wang, S. Tetrahedron 2005, 61, 3841.
- Lee, M. J.; Park, D. Y.; Lee, K. Y.; Kim, J. N. Tetrahedron Lett. 2006, 47, 1833.
- 163. Basavaiah, D.; Pandiaraju, S.; Krishnamacharyulu, M. Synlett 1996, 747.
- 164. Lee, C. G.; Lee, K. Y.; Lee, S.; Kim, J. N. Tetrahedron 2005, 61, 1493.
- Kim, S. C.; Gowrisankar, S.; Kim, J. N. Bull. Korean Chem. Soc. 2005, 26, 1001.
- 166. Pathak, R.; Madapa, S.; Batra, S. Tetrahedron 2007, 63, 451.
- (a) Masuyama, Y.; Nimura, Y.; Kurusu, Y. *Tetrahedron Lett.* 1991, 32, 225; (b) Choudhury, P. K.; Foubelo, F.; Yus, M. *Tetrahedron Lett.* 1998, 39, 3581; (c) Gonzalez, A. G.; Silva, M. H.; Padron, J. I.; Leon, F.; Reyes, E.; Alvarez-Mon, M.; Pivel, J. P.; Quintana, J.; Estevez, F.; Bermejo, J. *J. Med. Chem.* 2002, 45, 2358.
- Lee, K. Y.; Park, D. Y.; Kim, J. N. Bull. Korean Chem. Soc. 2006, 27, 1489
- Kabalka, G. W.; Venkataiah, B.; Chen, C. Tetrahedron Lett. 2006, 47, 4187.
- Lamer, A.-C. L.; Gouault, N.; David, M.; Boustie, J.; Uriac, P. J. Comb. Chem. 2006, 8, 643.
- 171. Ramachandran, P. V.; Pratihar, D. *Org. Lett.* **2007**, *9*, 4753.
- 172. Porto, R. S.; Coelho, F. Synth. Commun. 2006, 34, 3037.
- GowriSankar, S.; Lee, M. J.; Lee, S.; Kim, J. N. Bull. Korean Chem. Soc. 2004, 25, 1963.
- GowriSankar, S.; Lee, C. G.; Kim, J. N. Tetrahedron Lett. 2004, 45, 6949
- Lee, C. G.; Lee, K. Y.; Kim, S. J.; Kim, J. N. Bull. Korean Chem. Soc. 2007, 28, 719.
- 176. Lee, K. Y.; Lee, H. S.; Kim, J. N. Tetrahedron Lett. 2007, 48, 2007.
- 177. Eipert, M.; Maichle-Moessmer, C.; Maier, M. E. *Tetrahedron* 2003, 59, 7949.
- 178. Singh, V.; Batra, S. Synthesis 2006, 63.
- 179. Singh, V.; Batra, S., unpublished results.
- 180. Singh, V.; Madapa, S.; Batra, S. Synth. Commun., in press.
- 181. Kim, S. J.; Lee, H. S.; Kim, J. N. Tetrahedron Lett. 2007, 48, 1069.
- 182. Basavaiah, D.; Satyanarayana, T. Org. Lett. 2001, 3, 3619.
- 183. Kim, S. C.; Lee, H. S.; Kim, J. N. Bull. Korean Chem. Soc. 2007, 28, 147.
- 184. Nicholas, K. M.; O'Dell, D. K. Tetrahedron 2003, 59, 747.

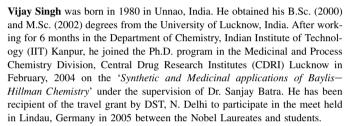
- 185. O'Dell, D. K.; Nicholas, K. M. J. Org. Chem. 2003, 68, 6427.
- 186. Lee, K. Y.; Lee, H. S.; Kim, J. N. Bull. Korean Chem. Soc. 2007, 28, 333
- Lee, K. Y.; Kim, J. M.; Kim, J. N. Bull. Korean Chem. Soc. 2002, 23, 1493.
- Basavaiah, D.; Mallikarjuna Reddy, R. M.; Kumaragurubaran, N.; Sharada. D. S. *Tetrahedron* 2002, 58, 3693.
- 189. Lee, K. Y.; Kim, J. M.; Kim, J. N. Tetrahedron 2003, 59, 385.
- Lee, K. Y.; Kim, S. C.; Kim, J. N. Bull. Korean Chem. Soc. 2005, 26, 1109.
- Amarante, G. W.; Benassi, M.; Sabino, A. A.; Esteves, P. M.; Coelho, F.;
 Eberlinb, M. N. Tetrahedron Lett. 2006, 47, 8427.
- 192. Basavaiah, D.; Rao, J. S.; Reddy, R. J. J. Org. Chem. 2004, 69, 7379.
- 193. Basavaiah, D.; Reddy, R. J.; Rao, J. S. Tetrahedron Lett. 2006, 47, 73.
- Shafiq, Z.; Liu, L.; Liu, Z.; Wang, D.; Chen, Y.-J. Org. Lett. 2007, 9, 2525.
- Familoni, O. B.; Klaas, P. J.; Lobb, K. A.; Pakade, V. E.; Kaye, P. T. Org. Biomol. Chem. 2006, 4, 3960.
- 196. Madapa, S.; Singh, V.; Batra, S. Tetrahedron 2006, 62, 8740.
- 197. Basavaiah, D.; Aravindu, K. Org. Lett. 2007, 9, 2453.
- 198. Basavaiah, D.; Rao, J. S. Tetrahedron Lett. 2004, 45, 1621.
- Lee, M. J.; Lee, K. Y.; Park, D. Y.; Kim, J. N. Bull. Korean Chem. Soc. 2005, 26, 1281.
- Lee, H. S.; Kim, S. J.; Kim, J. N. Bull. Korean Chem. Soc. 2006, 27, 1063.
- 201. Singh, V.; Kanojiya, S.; Batra, S. Tetrahedron 2006, 62, 10100.
- 202. Lee, K. Y.; Lee, Y. J.; Kim, J. N. Bull. Korean Chem. Soc. 2007, 28, 143.
- 203. Singh, V.; Batra, S. Eur. J. Org. Chem. 2007, 2970.
- Wasnaire, P.; Wiaux, M.; Touillaux, R.; Marko, I. E. Tetrahedron Lett.
 2006. 47, 985.
- 205. Shanmugam, P.; Vaithiyanathan, V.; Viswambharan, B. *Tetrahedron* **2006**, *62*, 4342.
- Singh, V.; Yadav, G. P.; Maulik, P. R.; Batra, S. Tetrahedron 2006, 62, 8731
- 207. Lee, M. J.; Kim, S. C.; Kim, J. N. Bull. Korean Chem. Soc. 2006, 27, 140.
- 208. Singh, V.; Pathak, R.; Batra, S. Catal. Commun. 2007, 8, 2054.
- Singh, V.; Yadav, G. P.; Maulik, P. R.; Batra, S. Tetrahedron 2008, 64, 2979.
- Kawahara, S.; Nakano, A.; Esumi, T.; Iwabuchi, Y.; Hatakeyama, S. Org. Lett. 2003, 5, 3103.
- Lee, K. Y.; Lee, C. G.; Kim, T. H.; Kim, J. N. Bull. Korean Chem. Soc. 2004, 25, 33.
- Pathak, R.; Roy, A. K.; Kanojiya, S.; Batra, S. Tetrahedron Lett. 2005, 46, 5289.
- 213. Pathak, R.; Roy, A. K.; Batra, S. Synlett 2005, 848.
- 214. Pathak, R.; Batra, S. Tetrahedron 2007, 63, 9448.
- 215. Pathak, R.; Batra, S.; Singh, N., unpublished results.
- Nag, S.; Pathak, R.; Kumar, M.; Shukla, P. K.; Batra, S. Bioorg. Med. Chem. Lett. 2006, 16, 3824.
- 217. Nag, S.; Yadav, G. P.; Maulik, P. R.; Batra, S. Synthesis 2007, 911.
- 218. Nag, S.; Madapa, S.; Batra, S. Synthesis 2008, 101.
- Kim, S. C.; Gowrisankar, S.; Kim, J. N. Tetrahedron Lett. 2006, 47, 3463.
- Lee, M. J.; Kim, S. C.; Kim, J. N. Bull. Korean Chem. Soc. 2006, 27, 439
- Kim, S. H.; Kim, K. H.; Kim, H. S.; Kim, J. N. Tetrahedron Lett. 2008, 49, 1948.
- 222. Lee, H. S.; Kim, J. M.; Kim, J. N. Tetrahedron Lett. 2007, 48, 4119.
- 223. Kim, J. M.; Lee, K. Y.; Kim, J. N. Bull. Korean Chem. Soc. 2002, 23, 1055.
- 224. Pathak, R.; Nag, S.; Batra, S. Synthesis 2006, 4205.
- 225. Batra, S.; Roy, A. K. Synthesis 2004, 2550.
- Back, T. G.; Rankic, D. A.; Sorbetti, J. M.; Wulff, J. E. Org. Lett. 2005, 7, 2377.
- Sorbetti, J. M.; Clary, K. N.; Rankic, D. A.; Wulff, J. E.; Parvez, M.;
 Back, T. G. J. Org. Chem. 2007, 72, 3326.
- 228. Tanaka, T.; Muto, T.; Maruoka, H.; Imajo, S.; Fukami, H.; Tomimori, Y.; Fukuda, Y.; Nakatsuka, T. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 3431.

- Maruoka, H.; Muto, T.; Tanaka, T.; Imajo, S.; Tomimori, Y.; Fukuda, Y.;
 Nakatsuka, T. Bioorg, Med. Chem. Lett. 2007, 17, 3435.
- Kim, S. J.; Kim, H. S.; Kim, T. H.; Kim, J. N. Bull. Korean Chem. Soc. 2007, 28, 1605.
- 231. Saxena, R.; Singh, V.; Batra, S. Tetrahedron 2004, 60, 10311.
- 232. Singh, V.; Saxena, R.; Batra, S. J. Org. Chem. 2005, 70, 353.
- Singh, V.; Madapa, S.; Yadav, G. P.; Maulik, P. R.; Batra, S. Synthesis 2006, 1995.
- Roy, A. K.; Pathak, R.; Yadav, G. P.; Maulik, P. R.; Batra, S. Synthesis 2006, 1021.
- 235. Saxena, R.; Batra, S. Poster at 9th Conference of Society of Chemists and Biologists, Rajkot, India, Jan 8–10, 2005.
- Kamimura, A.; Morita, R.; Matsuura, K.; Mitsudera, H.; Shirai, M. Tetrahedron 2003, 59, 9931.
- Coelho, F.; Veronese, D.; Lopes, E. C. S.; Rossi, R. C. *Tetrahedron Lett.* 2003, 44, 5731.
- Coelho, F.; Veronese, D.; Pavam, C. H.; de Paula, V. I.; Buffon, R. Tetrahedron 2006, 62, 4563.
- Kohn, L. K.; Pavam, C. H.; Veronese, D.; Coelho, F.; de Carvalho, J. E.;
 Almeida, W. P. Eur. J. Med. Chem. 2006, 41, 738.
- Kim, J. N.; Kim, J. M.; Lee, K. Y.; Gowrisankar, S. Bull. Korean Chem. Soc. 2004, 25, 1733.
- Gowrisankar, S.; Na, J. E.; Lee, M. J.; Kim, J. N. Bull. Korean Chem. Soc. 2005, 26, 319.
- Lee, K. Y.; Gowrisankar, S.; Lee, Y. J.; Kim, J. N. Tetrahedron 2006, 62, 8798.
- 243. Lee, K. Y.; Kim, J. M.; Kim, J. N. Tetrahedron Lett. 2003, 44, 6737.
- Mamaghani, M.; Tabatabaeian, K.; Mirzaeinejad, M.; Nikpassand, M. J. Iranian Chem. Soc. 2006, 3, 89.
- 245. Lee, K. Y.; Kim, S. C.; Kim, J. N. Bull. Korean Chem. Soc. 2006, 27, 319.
- 246. Zhang, Q.; Sun, S.; Hu, J.; Liu, Q.; Tan, J. J. Org. Chem. 2007, 72, 139.
- 247. Morizur, J.-F.; Mathias, L. J. *Tetrahedron Lett.* **2007**, *48*, 5555.
- 248. Lim, H. N.; Song, Y. S.; Lee, K.-J. Synthesis 2007, 3376.
- 249. Song, Y. S.; Lee, K.-J. Synthesis 2007, 3037.
- 250. Wasnaire, P.; de Merode, T.; Marko, I. E. Chem. Commun. 2007, 4755.
- 251. Ye, L.-W.; Han, X.; Sun, X.-L.; Tang, Y. Tetrahedron 2008, 64, 1487.
- Cabrera, S.; Aleman, J.; Bolze, P.; Bertelsen, S.; Jorgensen, K. A. Angew. Chem., Int. Ed. 2008, 47, 121.
- 253. Kim, H. S.; Kim, S. H.; Kim, J. N. Bull. Korean Chem. Soc. 2007, 28, 1841.
- Grange, R. L.; Ziogas, J.; North, A. J.; Angus, J. A.; Schiesser, C. H. Bioorg. Med. Chem. Lett. 2008, 18, 1241.
- Kim, J. M.; Kim, S. H.; Kim, J. N. Bull. Korean Chem. Soc. 2007, 28, 2505.
- Cha, M. J.; Song, Y. S.; Han, E.-G.; Lee, K.-J. J. Heterocycl. Chem. 2008, 235.
- Sa, M. M.; Fernandes, L.; Ferreira, M.; Bortoluzzi, A. J. *Tetrahedron Lett.* 2008, 49, 1228.
- (a) Vijaya Anand, R.; Baktharaman, S.; Singh, V. K. Tetrahedron Lett.
 2002, 43, 5393; (b) Loh, T.-P.; Cao, G.-Q.; Pei, J. Tetrahedron Lett.
 1998, 39, 1457; (c) Basavaiah, D.; Suguna Hyma, R. Tetrahedron
 1996, 52, 1253; (d) Grassi, D.; Lippuner, V.; Aebi, M.; Brunner, J.;
 Vasella, A. J. Am. Chem. Soc. 1997, 119, 10992; (e) Li, S.; Peng, L.;
 Zhang, T.; Li, Y. Gaodeng Xuexiao Huaxue Xuebao 2003, 24, 1410.
- Trost, B. M.; Thiel, O. R.; Tsui, H. C. J. Am. Chem. Soc. 2002, 124, 11616.

- Trost, B. M.; Thiel, O. R.; Tsui, H. C. J. Am. Chem. Soc. 2003, 125, 13155.
- Trost, B. M.; Machacek, M. R.; Tsui, H. C. J. Am. Chem. Soc. 2005, 127, 7014
- Krishna, P. R.; Narsingam, M.; Kannan, V. Tetrahedron Lett. 2004, 45, 4773
- 263. Almeida, W. P.; Coelho, F. Tetrahedron Lett. 2003, 44, 937.
- 264. De, G. P.; Silveira, C.; Coelho, F. Tetrahedron Lett. 2005, 46, 6477.
- Reddy, L. R.; Sarvanan, P.; Corey, E. J. J. Am. Chem. Soc. 2004, 126, 6230
- Reddy, L. R.; Fournier, J.-F.; Reddy, B. V. S.; Corey, E. J. Org. Lett. 2005, 7, 2699.
- Chen, C.-Y.; Chang, M.-Y.; Hsu, R.-T.; Chen, S.-T.; Chang, N.-C. Tetrahedron Lett. 2003, 44, 8627.
- Motozaki, T.; Sawamura, K.; Suzuki, A.; Yoshida, K.; Ueki, T.; Ohara,
 A.; Munakata, R.; Takao, K.-i.; Tadano, K.-i. Org. Lett. 2005, 7, 2265.
- Galeazzi, R.; Martelli, G.; Orena, M.; Rinaldia, S.; Sabatino, P. *Tetra-hedron* 2005, 61, 5465.
- Mergott, D. J.; Frank, S. A.; Roush, W. R. Proc. Natl. Acad. Sci. U.S.A. 2004, 101, 11955.
- 271. Seck, M.; Franck, X.; Seon-Meniel, B.; Hocquemiller, R.; Figadere, B. *Tetrahedron Lett.* **2006**, *47*, 4175.
- 272. Rodgen, S. A.; Schaus, S. E. Angew. Chem., Int. Ed. 2006, 45, 929.
- 273. Jogireddy, R.; Maier, M. E. J. Org. Chem. 2006, 71, 6999.
- Buu Hue, B. T.; Dijkink, J.; Kuiper, S.; van Schaik, S.; van Maarseveen,
 J. H.; Hiemstra, H. Eur. J. Org. Chem. 2006, 127.
- Nising, C. F.; Ohhemuller, U. K.; Brase, S. Angew. Chem., Int. Ed. 2006, 45, 307.
- Selvakumar, N.; Kumar, P. K.; Reddy, K. C. S.; Chary, B. C. *Tetrahedron Lett.* 2007, 48, 2021.
- Myers, E. L.; de Vries, J. G.; Aggarwal, V. K. Angew. Chem., Int. Ed. 2007, 46, 1893.
- Moore, C. G.; Murphy, P. J.; Williams, H. L.; McGown, A. T.; Smith, N. K. *Tetrahedron* 2007, 63, 11771.
- 279. Konig, C. M.; Harms, K.; Koert, U. Org. Lett. 2007, 9, 4777.
- Liang, H.; Schule, A.; Vors, J.-P.; Ciufolini, M. A. Org. Lett. 2007, 9, 4119
- 281. Das, B.; Banerjee, J.; Mahender, N.; Majhi, A. Org. Lett. 2004, 6, 3349.
- 282. Chapuis, C.; Büchi, G. H.; Wüest, H. Helv. Chim. Acta 2005, 88, 3069.
- Frezza, M.; Soulere, L.; Queneau, Y.; Doutheau, A. *Tetrahedron Lett.* 2005, 46, 6495.
- 284. Frezza, M.; Balestrino, D.; Soulère, L.; Reverchon, S.; Queneau, Y.; Forestier, C.; Doutheau, A. Eur. J. Org. Chem. 2006, 4731.
- Coelho, F.; Antonio, S.; Mateus, C. R.; Almeida, W. P. Braz. Pedido 2003003854, 2005; Chem. Abstr. 2006, 145, 397270.
- Das, B.; Chowdhury, N.; Banerjee, J.; Majhi, A. *Tetrahedron Lett.* 2006, 47, 6615.
- Das, B.; Banerjee, J.; Chowdhury, N.; Majhi, A.; Holla, H. Synlett 2006, 1879.
- 288. Das, B.; Banerjee, J.; Chowdhury, N.; Majhi, A.; Mahender, G. *Helv. Chim. Acta* **2006**, *89*, 876.
- Das, B.; Banerjee, J.; Chowdhury, N.; Majhi, A. Chem. Pharm. Bull. 2006, 54, 1725.
- Kang, S.-Y.; Park, K.-S.; Kim, J.; Chong, Y.; Choo, H. Bull. Korean Chem. Soc. 2007, 28, 179.
- 291. Kamal, A.; Krishnaji, T.; Reddy, P. V. Tetrahedron Lett. 2007, 48, 7232.

Biographical sketch







Sanjay Batra was born in 1965 in Lucknow, India. He did his graduation and post-graduation from DBS, College Dehradun. He completed his Ph.D. in 1993 under the supervision of Dr. A. P. Bhaduri in the Medicinal Chemistry Division at Central Drug Research Institute, Lucknow. He has been working as scientist in the Medicinal and Process Chemistry Division of CDRI since 1995. He also worked as Visiting Research Associate in the lab of Prof. M. A. Avery at the University of Mississippi at Oxford, USA. On his return in November 2001, he established his own research group at CDRI. His research interest includes development of chemistry associated with Baylis—Hillman reaction, heterocyclic, combinatorial, and medicinal chemistry.